=> d que 128 L1 STR Cy^G1 HO~B~OH H



VAR G1=4/7/12/18
REP G2=(1-4) C
NODE ATTRIBUTES:
CONNECT IS E1 RC AT 9
CONNECT IS E1 RC AT 10
CONNECT IS E1 RC AT 14
DEFAULT MLEVEL IS ATOM
GGCAT IS UNS AT 1
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 18

STEREO ATTRIBUTES: NONE

L3	13991	SEA FILE=REGISTRY SSS FUL	T.1	
L4		-	LU=ON	"LUMINESCENCE, CHEMILUMINESCEN
<u>1.4</u>	855/	CE"+OLD, NT/CT	,PO=ON	"LOMINESCENCE, CHEMILLOMINESCEN
L5	2835		LU=ON	"CHEMILUMINESCENCE SPECTROSCOP
ц	2033	Y"+OLD, NT/CT	D0-014	CHILITION TRADEDINGS STEETHOODS
L6	103	• •	LU=ON	"LUMINESCENCE, CHEMILUMINESCEN
		CE (L) DETECTORS"/CT		· ·
L7	8557	SEA FILE=HCAPLUS ABB=ON P	LU=ON	L6 OR L4
$^{\text{L8}}$	1055	SEA FILE=HCAPLUS ABB=ON P	LU=ON	"IMMUNOASSAY (L) CHEMILUMINESC
		ENCE"+OLD/CT		
L9	49		LU=ON	"GAS CHROMATOGRAPHIC DETECTORS
		(L) CHEMILUMINESCENCE"+OL	D/CT	
L10	938		LU=ON	CHEMILUMINESCENT SUBSTANCES+OL
		D/CT		
L11	28464	SEA FILE=HCAPLUS ABB=ON P	LU=ON	CHEMILUM?
L12	28962		LU=ON	(L4 OR L5 OR L6 OR L7 OR L8
		OR L9 OR L10 OR L11)		
L13	2461		LU=ON	"LUMINESCENCE, BIOLUMINESCENCE
		"+OLD,NT/CT		
L14	134		LU=ON	"IMMUNOASSAY (L) BIOLUMINESCEN
		CE"+OLD/CT		
L15	624		LU=ON	"LUMINESCENCE SPECTROSCOPY
		(L) BIOLUMINESCENCE"+OLD/C		
L16	181		LU=ON	"LUMINESCENT SUBSTANCES (L)
		BIOLUMINESCENT"+OLD/CT		
L17			LU=ON	BIOLUMIN?
L18	7004		LU=ON	(L13 OR L14 OR L15 OR L16 OR
		L17)		
L19	87681	SEA FILE=HCAPLUS ABB=ON P	LU=ON	FLUORESCENCE+NT/CT

```
22181 SEA FILE=HCAPLUS ABB=ON PLU=ON FLUORESCENT SUBSTANCES+OLD, NT/
L20
             393258 SEA FILE=HCAPLUS ABB=ON PLU=ON FLUORES?
L21
        393333 SEA FILE=HCAPLUS ABB=ON PLU=ON (L19 OR L20 OR L21)
L22
            36 SEA FILE=HCAPLUS ABB=ON PLU=ON L3 AND (L12 OR L18)
L24
             8 SEA FILE=REGISTRY ABB=ON PLU=ON L3 AND O2C2/ESS
L25
             2 SEA FILE=HCAPLUS ABB=ON PLU=ON L25
L26
             2 SEA FILE=HCAPLUS ABB=ON PLU=ON L26 AND (L12 OR L18 OR L22)
L27
            36 SEA FILE=HCAPLUS ABB=ON PLU=ON L24 OR L27
L28
```

=> d l28 ibib ab hitind hitstr 1-36

L28 ANSWER 1 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2004:722833 HCAPLUS

DOCUMENT NUMBER:

141:217988

TITLE:

Signalling compounds for use in methods of detecting

hydrogen peroxide

INVENTOR (S):

Akhavan-Tafti, Hashem; Eickholt, Robert A.; Lauwers,

Kenneth S.; Handley, Richard S.

PATENT ASSIGNEE(S):

SOURCE:

U.S. Pat. Appl. Publ., 33 pp., Cont.-in-part of U.S.

Ser. No. 371,053.

CODEN: USXXCO

DOCUMENT TYPE:

LANGUAGE:

Patent English

FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

	PAT	ENT 1	10.			KINI)	DATE			APPL	ICAT:	ION I	NO.	. 1	D	ATE	
•	US :	2004	17109	98		A1		2004								2	0030	520
	US :	2004	16653					2004									0030	
	WO :	2004						2004										
		W:	ΑE,	ΑE,	AG,	AL,	ΑL,	AM,	AM,	AM,	ΑT,	AT,	ΑU,	ΑZ,	AZ,	BA,	BB,	BG,
			BG,	BR,	BR,	BW,	BY,	BY,	BZ,	BZ,	CA,	CH,	CN,	CN,	CO,	CO,	CR,	CR,
			CU,	CU,	CZ,	CZ,	DE,	DE,	DK,	DK,	DM,	DZ,	EC,	EC,	EE,	EE,	EG,	ES,
								GE,										
			IS,	JP,	JP,	KE,	KE,	KG,	KG,	KP,	ΚP,	KP,	KR,	KR,	KZ,	KΖ,	ΚZ,	LC,
			LK,	LR,	LS,	LS,	LT,	LU,	LV,	MA,	MD,	MD,	MG,	MK,	MN,	MW,	MX,	MX,
					NA,													
		RW:	BW,	GH,	GM,	KE,	LS,	MW,	MZ,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AT,	BE,
								DK,										
			MC,	NL,	PT,	RO,	SE,	SI,	SK,	TR,	BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,
			GQ,	GW,	ML,	MR,	NE,	SN,	TD,	TG,	BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,
			GQ,	GW,	ML,	MR,	NE,	SN,	TD,	TG								
PRIO	RITY	APP	LN.	INFO	. :					1	US 2	003-3	3710	53	ì	A2 2	0030:	220
										1	US 2	003-	6009	28	7	A 2	0030	520

AB Compds. useful for detecting a source of hydrogen peroxide are disclosed wherein a signalling compound of the formula: Sig-B-(OR5)(OR6) is reacted with peroxide. Sig is an aromatic or heteroarom. ring group, B is a boron atom, and R5 and R6 are independently selected from hydrogen and lower alkyl groups and can be joined together as a straight or branched alkylene chain forming a five or six-membered ring. A detectable product compound Sig-OH Is produced and detected by measuring color, fluorescence, chemiluminescence, or bioluminescence. The signalling compound itself does not possess the detectable property or does so only to a very weak degree. The compds. can be used for detection in assays for peroxide or peroxide-producing enzymes and in assays employing

```
enzyme-labeled specific binding pairs.
     ICM C12Q001-26
IC
     ICS C07F005-02
NCL 435025000; 549213000
    79-3 (Inorganic Analytical Chemistry)
CC
     Section cross-reference(s): 7, 9
     Colorimetric indicators
IT
       Fluorescent indicators
      Luminescence, bioluminescence
      Luminescence, chemiluminescence
        (methods and compds. for detection of hydrogen peroxide)
IT
     741252-93-7
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (methods and compds. for detection of hydrogen peroxide)
     68572-88-3P 349666-24-6P 741252-95-9P
IT
     741252-96-0P 741252-97-1P
     RL: ARG (Analytical reagent use); PRP (Properties); RCT (Reactant); SPN
     (Synthetic preparation); ANST (Analytical study); PREP (Preparation); RACT
     (Reactant or reagent); USES (Uses)
        (methods and compds. for detection of hydrogen peroxide)
IT
     68572-87-2P, 9-Phenanthreneboronic Acid 709022-63-9P
     741252-90-4P 741252-91-5P 741252-92-6P
     741252-98-2P 741252-99-3P 741253-00-9P
     741253-01-0P 741253-02-1P 741253-04-3P
     741253-14-5P 741253-15-6P 741253-17-8P
     741253-18-9P
     RL: ARG (Analytical reagent use); PRP (Properties); SPN (Synthetic
     preparation); ANST (Analytical study); PREP (Preparation); USES (Uses)
        (methods and compds. for detection of hydrogen peroxide)
     164461-18-1P, 1-Pyrenylboronic acid 741253-05-4P
IT
     743459-14-5P
     RL: ARG (Analytical reagent use); SPN (Synthetic preparation); ANST
     (Analytical study); PREP (Preparation); USES (Uses)
        (methods and compds. for detection of hydrogen peroxide)
     129058-56-6P 134519-12-3P 741252-94-8P
IT
     RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (methods and compds. for detection of hydrogen peroxide)
                   741253-09-8P 741253-10-1P 741253-11-2P
     538338-71-5P
IT
     741253-12-3P ··
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (methods and compds. for detection of hydrogen peroxide)
     121445-45-2P 741253-06-5P 741253-07-6P
TT
     741253-08-7P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (methods and compds. for detection of hydrogen peroxide)
ΙT
    741252-93-7
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (methods and compds. for detection of hydrogen peroxide)
RN
     741252-93-7 HCAPLUS
     INDEX NAME NOT YET ASSIGNED
CN
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
     68572-88-3P 349666-24-6P 741252-95-9P
     741252-96-0P 741252-97-1P
     RL: ARG (Analytical reagent use); PRP (Properties); RCT (Reactant); SPN
     (Synthetic preparation); ANST (Analytical study); PREP (Preparation); RACT
     (Reactant or reagent); USES (Uses)
```

(methods and compds. for detection of hydrogen peroxide)

68572-88-3 HCAPLUS

RN1,3,2-Dioxaborolane, 4,4,5,5-tetramethyl-2-(9-phenanthrenyl)- (9CI) CNINDEX NAME)

349666-24-6 HCAPLUS RN

1,3,2-Dioxaborolane, 4,4,5,5-tetramethyl-2-(1-pyrenyl)- (9CI) (CA INDEX CNNAME)

741252-95-9 HCAPLUS RN

INDEX NAME NOT YET ASSIGNED CN

741252-96-0 HCAPLUS RN

CNINDEX NAME NOT YET ASSIGNED

$$\begin{array}{c} \text{Me} \\ \text{Me} \\ \text{Me} \end{array} \begin{array}{c} \text{O} \\ \text{B} \\ \text{O} \\ \text{N} \end{array} \begin{array}{c} \text{O} \\ \text{C-NH}_2 \\ \text{N} \end{array}$$

RN 741252-97-1 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 709022-63-9 HCAPLUS CN 1,3,2-Dioxaborolane, 2-(9-anthracenyl)-4,4,5,5-tetramethyl- (9CI) (CA INDEX NAME)

RN 741252-90-4 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741252-91-5 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741252-92-6 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741252-98-2 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741252-99-3 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741253-00-9 HCAPLUS CN Boronic acid, [2,2'-bibenzothiazol]-6-yl- (9CI) (CA INDEX NAME)

RN 741253-01-0 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741253-02-1 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741253-14-5 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741253-15-6 HCAPLUS

CN 4-Thiazolecarboxylic acid, 2-(6-borono-2-benzothiazolyl)-4,5-dihydro-5,5-dimethyl- (9CI) (CA INDEX NAME)

$$\begin{array}{c|c} \text{OH} & \\ \mid & \\ \text{HO}-\text{ B} & \\ \hline \mid & \\ \text{N} & \\ \text{S} & \\ \text{Me} & \\ \end{array}$$

RN 741253-17-8 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741253-18-9 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

IT 164461-18-1P, 1-Pyrenylboronic acid 741253-05-4P 743459-14-5P

RL: ARG (Analytical reagent use); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses) (methods and compds. for detection of hydrogen peroxide)

RN 164461-18-1 HCAPLUS

CN Boronic acid, 1-pyrenyl- (9CI) (CA INDEX NAME)

RN 741253-05-4 HCAPLUS CN Boronic acid, 1H-indol-3-yl- (9CI) (CA INDEX NAME)

RN 743459-14-5 HCAPLUS

CN INDEX NAME NOT YET ASSIGNED

IT 741252-94-8P

RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(methods and compds. for detection of hydrogen peroxide)

RN 741252-94-8 HCAPLUS

CN INDEX NAME NOT YET ASSIGNED

IT 741253-12-3P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(methods and compds. for detection of hydrogen peroxide)

RN 741253-12-3 HCAPLUS

CN Boronic acid, [3-[4-(1,1-dimethylethyl)tetrahydro-5,5-dimethyl-3-furanyl]phenyl]- (9CI) (CA INDEX NAME)

IT 741253-06-5P 741253-07-6P 741253-08-7P

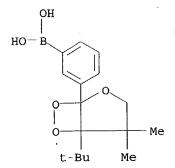
RL: SPN (Synthetic preparation); PREP (Preparation) (methods and compds. for detection of hydrogen peroxide)

RN 741253-06-5 HCAPLUS

CN INDEX NAME NOT YET ASSIGNED

RN 741253-07-6 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741253-08-7 HCAPLUS
CN INDEX NAME NOT YET ASSIGNED



L28 ANSWER 2 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2004:701711 HCAPLUS

DOCUMENT NUMBER:

141:217987

TITLE:

Signalling compounds and methods for detecting

hydrogen peroxide

INVENTOR(S):

Akhavan-Tafti, Hashem; Eickholt, Robert A.; Lauwers,

Kenneth S.; Handley, Richard S.

PATENT ASSIGNEE(S):

SOURCE:

U.S. Pat. Appl. Publ., 32 pp.

CODEN: USXXCO

DOCUMENT TYPE:

LANGUAGE:

Patent English

FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

PA	TENT	NO.		=	KIN	D	DATE		j	APPL	ICAT:	ION I	. O <i>V</i>		D	ATE	
						-									-		
US	2004	1665	39		A1		2004	0826	1	US 2	003-3	3710	53		2	0030:	220
ບຣ	2004	1710	98		A1		2004	0902	1	US 2	003-	5009:	28	•	2	0030	520
WO	2004	0748	10		A2		2004	0902	1	WO 2	004-1	JS20	20		2	0040	217
	W:	AE,	ΑE,	AG,	AL,	AL,	AM,	AM,	AM,	AT,	ΑT,	ΑU,	ΑZ,	ΑZ,	BA,	BB,	BG,
		BG,	BR,	BR,	BW,	BY,	BY,	BZ,	BZ,	CA,	CH,	CN,	CN,	CO,	CO,	CR,	CR,
							DE,										
		ES,	FI,	FI,	GB,	GD,	GE,	GE,	GH,	GM,	HR,	HR,	HU,	HU,	ID,	IL,	IN,
		ıs,	JP,	JP,	KE,	KΕ,	KG,	KG,	ΚP,	KP,	ΚP,	KR,	KR,	KZ,	KΖ,	ΚZ,	LC,
		LK,	LR,	LS,	LS,	LT,	LU,	LV,	MA,	MD,	MD,	MG,	MK,	MN,	MW,	MX,	MX,
		MZ,	MZ,	NA,	NI												
	RW:	BW,	GH,	GM,	KE,	LS,	MW,	MZ,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AT,	BE,
		BG,	CH,	CY,	CZ,	DE,	DK,	EE,	ES,	FI,	FR,	GB,	GR,	HU,	IE,	IT,	LU,
							SI,										
							SN,										
							SN,										
PRIORIT	Y APP				•	•		•		US 2	003-	3710	53		A2 2	0030	220
111101111										US 2	003-	6009	28		A 2	0030	620
											_						

Methods and compound useful for detecting a source of hydrogen peroxide are disclosed wherein a signalling compound of the formula: Sig-B-(OR)2 is reacted with peroxide. Sig is a non-polymeric organic group, B is a boron atom, and each R is independently selected from hydrogen, alkyl and aryl groups and can be joined together as a straight or branched alkylene chain forming a ring or as an aromatic ring. A detectable product compound Sig-OH or Sig-O - is produced and detected by measuring color, absorbance,

fluorescence, chemiluminescence, or

bioluminescence. The signalling compound itself does not possess the detectable property or does so only to a very weak degree. The

```
methods can be used as a detectable signal in assays for peroxide or
     peroxide-producing enzymes and in assays employing enzyme-labeled specific
     binding pairs.
    ICM G01N033-53
IC
    435007900
NCL
     79-3 (Inorganic Analytical Chemistry)
CC
     Section cross-reference(s): 7, 9
    Colorimetric indicators
TT
       Fluorescent indicators
       Luminescence, bioluminescence
       Luminescence, chemiluminescence
        (methods and compds. for detection of hydrogen peroxide)
IT
     741252-93-7
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (methods and compds. for detection of hydrogen peroxide)
     68572-88-3P 349666-24-6P, 4,4,5,5-Tetramethyl-2-(1-
TT
     pyrenyl) -1,3,2-dioxaborolane 741252-95-9P 741252-96-0P
     741252-97-1P
     RL: ARG (Analytical reagent use); PRP (Properties); RCT (Reactant); SPN
     (Synthetic preparation); ANST (Analytical study); PREP (Preparation); RACT
     (Reactant or reagent); USES (Uses)
        (methods and compds. for detection of hydrogen peroxide)
     68572-87-2P, 9-Phenanthreneboronic Acid 709022-63-9P
IT
     741252-90-4P 741252-91-5P 741252-92-6P
     741252-98-2P 741252-99-3P 741253-00-9P
     741253-01-0P 741253-02-1P 741253-04-3P
     741253-14-5P 741253-15-6P 741253-17-8P
     741253-18-9P
     RL: ARG (Analytical reagent use); PRP (Properties); SPN (Synthetic
     preparation); ANST (Analytical study); PREP (Preparation); USES (Uses)
        (methods and compds. for detection of hydrogen peroxide)
IT
     164461-18-1P, 1-Pyrenylboronic acid 741253-05-4P
     RL: ARG (Analytical reagent use); SPN (Synthetic preparation); ANST
     (Analytical study); PREP (Preparation); USES (Uses)
        (methods and compds. for detection of hydrogen peroxide)
                                                                 134519-12-3P
     129058-56-6P, Methyl 6-hydroxybenzothiazole-2-carboxylate
IT
     RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (methods and compds. for detection of hydrogen peroxide)
                                                 741253-11-2P
                    741253-09-8P 741253-10-1P
IT
     538338-71-5P
     741253-12-3P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (methods and compds. for detection of hydrogen peroxide)
     121445-45-2P 741253-06-5P 741253-07-6P
TT
     741253-08-7P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (methods and compds. for detection of hydrogen peroxide)
IT
     741252-93-7
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (methods and compds. for detection of hydrogen peroxide)
     741252-93-7 HCAPLUS
RN
     INDEX NAME NOT YET ASSIGNED
CN
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
     68572-88-3P 349666-24-6P, 4,4,5,5-Tetramethyl-2-(1-
     pyrenyl) -1,3,2-dioxaborolane 741252-95-9P 741252-96-0P
     741252-97-1P
```

RL: ARG (Analytical reagent use); PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)

(methods and compds. for detection of hydrogen peroxide)

RN 68572-88-3 HCAPLUS

CN 1,3,2-Dioxaborolane, 4,4,5,5-tetramethyl-2-(9-phenanthrenyl)- (9CI) (CA INDEX NAME)

RN 349666-24-6 HCAPLUS CN 1,3,2-Dioxaborolane, 4,4,5,5-tetramethyl-2-(1-pyrenyl)- (9CI) (CA INDEX NAME)

RN 741252-95-9 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741252-96-0 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741252-97-1 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

IT 68572-87-2P, 9-Phenanthreneboronic Acid 709022-63-9P
741252-90-4P 741252-91-5P 741252-92-6P
741252-98-2P 741252-99-3P 741253-00-9P
741253-01-0P 741253-02-1P 741253-04-3P
741253-14-5P 741253-15-6P 741253-17-8P
741253-18-9P
RL: ARG (Analytical reagent use); PRP (Properties); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses) (methods and compds. for detection of hydrogen peroxide)
RN 68572-87-2 HCAPLUS
CN Boronic acid, 9-phenanthrenyl- (9CI) (CA INDEX NAME)

RN 709022-63-9 HCAPLUS
CN 1,3,2-Dioxaborolane, 2-(9-anthracenyl)-4,4,5,5-tetramethyl- (9CI) (CA
INDEX NAME)

RN 741252-90-4 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741252-91-5 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741252-92-6 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741252-98-2 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741252-99-3 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741253-00-9 HCAPLUS CN Boronic acid, [2,2'-bibenzothiazol]-6-yl- (9CI) (CA INDEX NAME)

RN 741253-01-0 HCAPLUS

CN

INDEX NAME NOT YET ASSIGNED

RN 741253-02-1 HCAPLUS

CN INDEX NAME NOT YET ASSIGNED

RN 741253-04-3 HCAPLUS

CN 4-Thiazolecarboxylic acid, 2-(6-borono-2-benzothiazolyl)- (9CI) (CA INDEX NAME)

RN 741253-14-5 HCAPLUS

CN INDEX NAME NOT YET ASSIGNED

RN 741253-15-6 HCAPLUS

CN 4-Thiazolecarboxylic acid, 2-(6-borono-2-benzothiazolyl)-4,5-dihydro-5,5-dimethyl- (9CI) (CA INDEX NAME)

HO-B
$$S$$
 N
 S
 Me
 Me

RN 741253-17-8 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741253-18-9 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741253-05-4 HCAPLUS CN Boronic acid, 1H-indol-3-yl- (9CI) (CA INDEX NAME)

IT 741252-94-8P

RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (methods and compds. for detection of hydrogen peroxide)

RN 741252-94-8 HCAPLUS

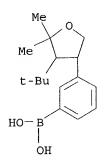
CN INDEX NAME NOT YET ASSIGNED

IT 741253-12-3P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (methods and compds. for detection of hydrogen peroxide)

RN 741253-12-3 HCAPLUS

CN Boronic acid, [3-[4-(1,1-dimethylethyl)tetrahydro-5,5-dimethyl-3-furanyl]phenyl]- (9CI) (CA INDEX NAME)



IT 741253-06-5P 741253-07-6P 741253-08-7P

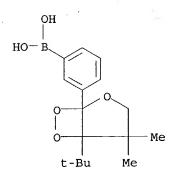
RL: SPN (Synthetic preparation); PREP (Preparation) (methods and compds. for detection of hydrogen peroxide)

RN 741253-06-5 HCAPLUS

CN INDEX NAME NOT YET ASSIGNED

RN 741253-07-6 HCAPLUS CN INDEX NAME NOT YET ASSIGNED

RN 741253-08-7 HCAPLUS CN INDEX NAME NOT YET ASSIGNED



L28 ANSWER 3 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2004:294829 HCAPLUS

DOCUMENT NUMBER: TITLE:

141:6779

Well-divided and pH-dependent bimodal

chemiluminescence of 2-methyl-6-phenyl-8-(4-

substituted phenyl)-imidazo[1,2-a]pyrazin-3(7H)-ones

induced by superoxide anion

AUTHOR(S):

Saito, Ryota; Inoue, Chizuru; Katoh, Akira

CORPORATE SOURCE:

Department of Applied Chemistry, Seikei University,

Tokyo, 180-8633, Japan

SOURCE:

Heterocycles (2004), 63(4), 759-764

```
CODEN: HTCYAM; ISSN: 0385-5414 Broken
                         Japan Institute of Heterocyclic Chemistry
PUBLISHER:
DOCUMENT TYPE:
                         Journal
LANGUAGE:
                         English
     An unprecedented pH-dependent bimodal chemiluminescence of
     2-methyl-6-phenyl-8-(4-substituted phenyl)imidazo[1,2-a]pyrazin-3(7H)-ones
     (4-substituent = CF3 for a, H for b) initiated by superoxide anion
     (O2◆-) in phosphate buffer solns. is described. The intensity ratio
     of the bimodal luminescence due to two light-emitting species, the
     singlet-excited neutral 2-acetamido-5-phenyl-3-(4-substituted
     phenyl)pyrazine and its amide anion, varied as the pH rose. The
     percentage of the anionic luminescence intensity
     [[Ianion/(Ineutral+Ianion)] for a showed a good linear correlation with the
     pH value, demonstrating its usefulness as a pH indicator as well as an
     02 -- probe.
     22-9 (Physical Organic Chemistry)
CC
     Section cross-reference(s): 9
IT
     Substituent effects
        (electronic, on chemiluminescence; well-separated and
        pH-dependent bimodal chemiluminescence of
        2-methyl-6-phenyl-8-(4-substituted phenyl)imidazo[1,2-a]pyrazin-3(7H)-
        ones induced by superoxide anion)
     Bathochromic effect
IT
        (in anion chemiluminescence vs. neutral; well-separated and
        pH-dependent bimodal chemiluminescence of
        2-methyl-6-phenyl-8-(4-substituted phenyl)imidazo[1,2-a]pyrazin-3(7H)-
        ones induced by superoxide anion)
     Acid-base indicators
TT
       Luminescence, chemiluminescence
        (well-separated and pH-dependent bimodal chemiluminescence of
        2-methyl-6-phenyl-8-(4-substituted phenyl)imidazo[1,2-a]pyrazin-3(7H)-
        ones induced by superoxide anion)
     98-80-6, Phenylboronic acid
                                   24241-18-7, 2-Amino-3,5-
IT
     dibromopyrazine
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (Suzuki coupling; well-separated and pH-dependent bimodal
        chemiluminescence of 2-methyl-6-phenyl-8-(4-substituted
        phenyl) imidazo[1,2-a]pyrazin-3(7H) -ones induced by superoxide anion)
     625848-14-8P
IT
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (Suzuki coupling; well-separated and pH-dependent bimodal
        chemiluminescence of 2-methyl-6-phenyl-8-(4-substituted
        phenyl)imidazo[1,2-a]pyrazin-3(7H)-ones induced by superoxide anion)
     694495-36-8
                   694495-37-9
IT
     RL: FMU (Formation, unclassified); PRP (Properties); FORM (Formation,
     nonpreparative)
        (anionic emitting species; well-separated and pH-dependent bimodal
        chemiluminescence of 2-methyl-6-phenyl-8-(4-substituted
        phenyl)imidazo[1,2-a]pyrazin-3(7H)-ones induced by superoxide anion)
IT
     41270-70-6P
                   694495-32-4P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (heterocyclization; well-separated and pH-dependent bimodal
        chemiluminescence of 2-methyl-6-phenyl-8-(4-substituted
        phenyl)imidazo[1,2-a]pyrazin-3(7H)-ones induced by superoxide anion)
                  694495-35-7
IT
     74152-22-0
     RL: FMU (Formation, unclassified); PRP (Properties); FORM (Formation,
     nonpreparative)
```

(neutral emitting species; well-separated and pH-dependent bimodal chemiluminescence of 2-methyl-6-phenyl-8-(4-substituted phenyl)imidazo[1,2-a]pyrazin-3(7H)-ones induced by superoxide anion) 694495-34-6P IT 694495-33-5P RL: ARG (Analytical reagent use); RCT (Reactant); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); RACT (Reactant or reagent); USES (Uses) (target indicator; well-separated and pH-dependent bimodal chemiluminescence of 2-methyl-6-phenyl-8-(4-substituted phenyl)imidazo[1,2-a]pyrazin-3(7H)-ones induced by superoxide anion) 11062-77-4, Superoxide IT RL: ANT (Analyte); RCT (Reactant); ANST (Analytical study); RACT (Reactant or reagent) (well-separated and pH-dependent bimodal chemiluminescence of 2-methyl-6-phenyl-8-(4-substituted phenyl)imidazo[1,2-a]pyrazin-3(7H)ones induced by superoxide anion) 625848-24-0P IT RL: BYP (Byproduct); PREP (Preparation) (well-separated and pH-dependent bimodal chemiluminescence of 2-methyl-6-phenyl-8-(4-substituted phenyl)imidazo[1,2-a]pyrazin-3(7H)ones induced by superoxide anion) 78-98-8, Pyruvaldehyde 128796-39-4, 4-IT (Trifluoromethyl) phenylboronic acid RL: RCT (Reactant); RACT (Reactant or reagent) (well-separated and pH-dependent bimodal chemiluminescence of 2-methyl-6-phenyl-8-(4-substituted phenyl)imidazo[1,2-a]pyrazin-3(7H)ones induced by superoxide anion) 98-80-6, Phenylboronic acid IT RL: RCT (Reactant); RACT (Reactant or reagent) (Suzuki coupling; well-separated and pH-dependent bimodal chemiluminescence of 2-methyl-6-phenyl-8-(4-substituted phenyl)imidazo[1,2-a]pyrazin-3(7H)-ones induced by superoxide anion) 98-80-6 HCAPLUS RN Boronic acid, phenyl- (9CI) (CA INDEX NAME) CNPh HO- B- OH 128796-39-4, 4-(Trifluoromethyl) phenylboronic acid TT RL: RCT (Reactant); RACT (Reactant or reagent) (well-separated and pH-dependent bimodal chemiluminescence of 2-methyl-6-phenyl-8-(4-substituted phenyl)imidazo[1,2-a]pyrazin-3(7H)ones induced by superoxide anion) 128796-39-4 HCAPLUS RN Boronic acid, [4-(trifluoromethyl)phenyl]- (9CI) (CA INDEX NAME) CN

REFERENCE COUNT: 16 THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 4 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2003:981793 HCAPLUS

DOCUMENT NUMBER:

141:25070

TITLE:

Chemiluminescent properties of blue fluorophores containing naphthalene unit

AUTHOR(S):

Cheon, Jong-Woo; Lee, Chil-Won; Gong, Myoung-Seon;

Geum, Neri

CORPORATE SOURCE:

Department of Chemistry, Dankook University, Chungnam,

330-714, S. Korea

SOURCE:

Dyes and Pigments (2004), 61(1), 23-30

CODEN: DYPIDX; ISSN: 0143-7208

PUBLISHER:

Elsevier Science Ltd.

DOCUMENT TYPE: LANGUAGE:

Journal English

OTHER SOURCE(S):

CASREACT 141:25070

Various conjugated blue fluorophores containing naphthalene moiety in biphenyl analogs were synthesized via Suzuki reaction for the blue chemiluminescent fluorophore. UV-Vis absorption, photoluminescence and chemiluminescence were measured and evaluated. The fluorophores displayed blue photoluminescence in solution with maximum around 375-410 nm. Sodium salicylate-catalyzed reaction of hydrogen peroxide with bis(2-carbopentyloxy-3,5,6-trichlorophenyl) oxalate (CPPO) produced a strong chemiluminescent blue light emission with wavelengths of 398-420 nm in the presence of the fluorophores. The chemiluminescent intensity decayed exponentially and the glow of chemiluminescence maintained for more than 4 h and was visible with naked eye.

CC 41-2 (Dyes, Organic Pigments, Fluorescent Brighteners, and Photographic Sensitizers)

ST chemiluminescence naphthalene deriv blue chromophore

IT Luminescence

Luminescence, chemiluminescence

(chemiluminescent properties of blue fluorophores containing naphthalene unit)

IT Molecular structure-property relationship

(luminescence; chemiluminescent properties of blue fluorophores containing naphthalene unit)

IT 75203-51-9

RL: NUU (Other use, unclassified); USES (Uses)

(chemiluminescent properties of blue fluorophores containing naphthalene unit)

IT 96589-92-3P 142450-41-7P 699008-32-7P 699008-33-8P 699008-34-9P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (chemiluminescent properties of blue fluorophores containing

naphthalene unit)

IT 92-86-4, 4,4'-Dibromobiphenyl 13922-41-3, 1-Naphthaleneboronic acid 32316-92-0, 2-Naphthaleneboronic acid 121602-03-7

156641-98-4 188200-91-1

RL: RCT (Reactant); RACT (Reactant or reagent)

(chemiluminescent properties of blue fluorophores containing naphthalene unit)

IT 13922-41-3, 1-Naphthaleneboronic acid 32316-92-0,

2-Naphthaleneboronic acid 156641-98-4

RL: RCT (Reactant); RACT (Reactant or reagent)

(chemiluminescent properties of blue fluorophores containing

naphthalene unit)

13922-41-3 HCAPLUS RN

CN. Boronic acid, 1-naphthalenyl- (9CI) (CA INDEX NAME)

32316-92-0 HCAPLUS RN

Boronic acid, 2-naphthalenyl- (9CI) (CA INDEX NAME) CN

156641-98-4 HCAPLUS RN

Boronic acid, (6-methoxy-2-naphthalenyl)- (9CI) (CA INDEX NAME) CN

REFERENCE COUNT:

THERE ARE 25 CITED REFERENCES AVAILABLE FOR THIS 25 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 5 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2003:746761 HCAPLUS

DOCUMENT NUMBER:

139:395735

TITLE:

Synthesis of 3,7-dihydroimidazo[1,2a]pyrazine-3-ones

and their chemiluminescent properties

AUTHOR (S):

Adamczyk, Maciej; Akireddy, Srinivasa Rao; Johnson,

Donald D.; Mattingly, Phillip G.; Pan, You; Reddy,

Rajarathnam E.

CORPORATE SOURCE:

Department of Chemistry, Diagnostics Division, Abbott

Laboratories, Abbott Park, IL, 60064-6016, USA

SOURCE:

Tetrahedron (2003), 59(41), 8129-8142

CODEN: TETRAB; ISSN: 0040-4020

PUBLISHER:

Elsevier Science B.V.

DOCUMENT TYPE:

Journal

```
LANGUAGE:
                         English
CASREACT 139:395735
                         Enalish
OTHER SOURCE(S):
     A series of 3,7-dihydroimidazo[1,2a]pyrazine-3-ones, e.g., I, were prepared
     from 2-amino-3,5-dibromopyrazine. The concise synthesis of coelenterazine
     I, in three steps, 48% overall yield and >99% purity exemplifies the
     strategy. Further, the synthetic approach facilitated the regiospecific
     incorporation of carboxyalkyl linkers on the 3,7-
     dihydroimidazo[1,2a]pyrazine-3-one nucleus that are required for
     bioconjugation. Peroxymonocarbonate, an electrophilic oxidant, was used
     to initiate pseudo-flash' chemiluminescence from this class of
     mols.
CC
     26-9 (Biomolecules and Their Synthetic Analogs)
     Section cross-reference(s): 28
     aminodibromopyrazine arylboronic acid Suzuki coupling palladium;
ST
     aminoaralkylbromopyrazine prepn; arylboronic acid
     aminoaralkylbromopyrazine Suzuki coupling palladium; amino substituted
     pyrazine prepn; pyrazinamine ketoacetal condensation; imidazopyrazinone
     prepn chemiluminescence; pyrazinone imidazo prepn
     chemiluminescence; coelenterazine total synthesis
     chemiluminescence
     Luminescence, chemiluminescence
IT
        (preparation and chemiluminescent properties of
        dihydroimidazo[1,2a]pyrazinones via cyclocondensation of
        aminodisubstituted pyrazines with \alpha-ketoacetals)
                    625848-46-6P
                                   625848-47-7P
                                                  625848-48-8P
                                                                 625848-49-9P
IT
     625848-45-5P
                    625848-51-3P
                                   625848-52-4P
                                                  625848-53-5P
                                                                 625848-54-6P
     625848-50-2P
                                   625848-57-9P
     625848-55-7P
                    625848-56-8P
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (preparation and chemiluminescent properties of
        dihydroimidazo[1,2a]pyrazinones via cyclocondensation of
        aminodisubstituted pyrazines with \alpha-ketoacetals)
                                   24241-18-7,
     98-80-6 1765-93-1 5720-05-8
IT
     2-Amino-3,5-dibromopyrazine 128796-39-4 159191-56-7
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (preparation of aminoarylpyrazines via palladium catalyzed Suzuki coupling
        of aminodibromopyrazine with arylboronic acids)
     98-80-6 1765-93-1 5720-05-8
IT
     128796-39-4 159191-56-7
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (preparation of aminoarylpyrazines via palladium catalyzed Suzuki coupling
        of aminodibromopyrazine with arylboronic acids)
RN
     98-80-6 HCAPLUS
     Boronic acid, phenyl- (9CI) (CA INDEX NAME)
CN
   Ph
HO- B- OH
```

Boronic acid, (4-fluorophenyl) - (9CI) (CA INDEX NAME)

RN

CN

1765-93-1 HCAPLUS

RN 5720-05-8 HCAPLUS

CN Boronic acid, (4-methylphenyl) - (9CI) (CA INDEX NAME)

RN 128796-39-4 HCAPLUS

CN Boronic acid, [4-(trifluoromethyl)phenyl] - (9CI) (CA INDEX NAME)

RN 159191-56-7 HCAPLUS

CN Boronic acid, [4-[[(1,1-dimethylethyl)dimethylsilyl]oxy]phenyl]- (9CI) (CA INDEX NAME)

REFERENCE COUNT:

47 THERE ARE 47 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 6 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2003:617590 HCAPLUS

DOCUMENT NUMBER:

139:292201

TITLE:

Diversity Synthesis via C-H Bond Functionalization:

```
Concept-Guided Development of New C-Arylation Methods
                         for Imidazoles
                         Sezen, Bengue; Sames, Dalibor
AUTHOR(S):
                         Department of Chemistry, Columbia University, New
CORPORATE SOURCE:
                         York, NY, 10027, USA
                         Journal of the American Chemical Society (2003),
SOURCE:
                         125(35), 10580-10585
                         CODEN: JACSAT; ISSN: 0002-7863
                         American Chemical Society
PUBLISHER:
                         Journal
DOCUMENT TYPE:
                         English
LANGUAGE:
                         CASREACT 139:292201
OTHER SOURCE(S):
    Herein, the concept of systematic derivatization of a structural motif via
     C-H bond functionalization was formulated. This concept may not only
     serve as a blueprint for new strategies in diversity synthesis but also
     provide systematic guidance for the identification of unsolved and
     important synthetic challenges. To illustrate this point,
     2-phenylimidazole was selected as the core motif for this study, a choice
     inspired by numerous azole-based synthetics, including pharmaceuticals
     (compound SB 202190), and also fluorescent and chemiluminescent
     probes. It was possible to show that systematic and comprehensive
     arylation of the 2-phenylimidazole core was feasible, and in the context
     of this study new arylation methods were developed. The direct
     4-arylation of free 2-phenylimidazole was achieved with iodoarenes as the
     aryl donors in the presence of palladium catalyst (Pd/Ph3P) and magnesium
     oxide as the base. A complete switch from C-4 to C-2' arylation was
     accomplished using a ruthenium catalyst [CpRu(Ph3P)2Cl] and Cs2CO3. The
     corresponding transformations for (N,2)-diphenylimidazole (C-5 and C-2'
     arylation) were accomplished via the palladium-based method
     [Pd(OAc)2/Ph3P/Cs2CO3] and a rhodium-catalyzed procedure
     [Rh(acac)(CO)2/Cs2CO3], resp. All of the arylation methods described
     herein demonstrated broad synthetic scope, high efficiency, and exclusive
     selectivity. Furthermore, these new methods proved to be orthogonal to one
     another and applicable to sequential arylation schemes. With these
     methods in hand, arrays of arylated imidazoles may now be accessed in a
     direct manner from 2-phenylimidazole. This strategy stands in sharp
     contrast to a traditional approach, wherein a distinct and multistep
     synthesis would be required for each analog.
     28-9 (Heterocyclic Compounds (More Than One Hetero Atom))
     Section cross-reference(s): 25
                                      104-92-7, 1-Bromo-4-
     98-80-6, Boronic acid, phenyl-
IT
     methoxybenzene 108-86-1, Bromobenzene, reactions 371-40-4,
                          402-43-7, 1-Bromo-4-(trifluoromethyl)benzene
     4-Fluorobenzenamine
     403-46-3, 4-Fluoro-N,N-dimethylaniline 455-14-1, 4-
(Trifluoromethyl)benzenamine 591-50-4, Iodobenzene
     1-Iodo-2-methylbenzene
                              670-96-2, 2-Phenyl-1H-imidazole
                                                                 696-62-8,
     1-Iodo-4-methoxybenzene
                               1120-87-2, 4-Bromopyridine 13329-40-3,
     1-(4-Iodophenyl)ethanone
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (diversity synthesis via carbon-hydrogen bond functionalization;
        concept-guided development of carbon arylation methods for imidazoles)
     98-80-6, Boronic acid, phenyl-
TT
     RL: RCT (Reactant); RACT (Reactant or reagent)
```

(diversity synthesis via carbon-hydrogen bond functionalization; concept-guided development of carbon arylation methods for imidazoles)

98-80-6 HCAPLUS

Boronic acid, phenyl- (9CI) (CA INDEX NAME)

RN

CN

```
Ph
|
HO-- B-- OH
```

REFERENCE COUNT:

THERE ARE 35 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 7 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2003:319853 HCAPLUS

DOCUMENT NUMBER:

138:337834

TITLE:

Preparation of aryl-β-diketones as luminous compounds and labeling reagents using the same

INVENTOR (S):

Saito, Michihiro; Pretsch, Ernoe

PATENT ASSIGNEE(S):

Hitachi High Technologies Corporation, Japan

SOURCE:

PCT Int. Appl., 51 pp. CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND DATE	APPLICATION NO.	DATE
WO 2003033447	A1 20030424	WO 2002-JP10511	20021010
W: CN, JP, KR,	US		
RW: AT, BE, BG,	CH, CY, CZ, DE,	DK, EE, ES, FI, FR, G	B, GR, IE, IT,
	PT, SE, SK, TR		•
		EP 2002-772986	
		GB, GR, IT, LI, LU, N	
IE, SI, LT,	LV, FI, RO, MK,	CY, AL, TR, BG, CZ, E	E, SK
PRIORITY APPLN. INFO .:		JP 2001-312562	A 20011010
		WO 2002-JP10511	W 20021010

OTHER SOURCE(S): MARPAT 138:337834

Disclosed are compds. represented by the following general formula R-Y-(-X-Phe-COCH2COCnF2n+1)m (wherein R represents hydrogen, alkyl, Ph or a group capable of binding to a protein, a peptide, an amino acid, a nucleic acid or a base; Y represents CH2, a carbon ring or a heterocycle; X represents O, S, NH, CH2, OCH2, CONH or NHCO; Phe represents phenylene; n is an integer of from 1 to 5; and m is 1, 2 or 3), luminous complexes comprising the above compds. with rare earth ions, labeling reagents comprising the compds. or luminous complexes as described above, and a method of labeling a protein, a peptide, an amino acid, a nucleic acid or a base using the above labeling reagents. When these compds. are complexed with metal ions, they emit fluorescence, delayed fluorescence, or phosphorescence and are suitable as labeling agents for time-resolved fluorometry, delayed phosphorimetry, or energy-transfer fluorometry used in nucleic acid detection, immunoassay, or chemiluminescent method. Thus, a mixture of 1,2-bis(bromomethyl)benzene 5.0, 4-acetylphenylboronic acid 13.6, CsCO3 18.5 g, 50 mL THF, and 5 mL H2O was stirred at 70° for 30 min, treated with 1.5 g PdCl2(dppf).CH2Cl2 [dppf = 1,1'-bis(diphenylphosphino) ferrocene], and heated for 24 h to give, after workup and silica gel chromatog., 15% 1,2-bis(4acetylbenzyl)benzene (I). I 300, C3F7CO2Et 440, NaOMe 99 mg, and 12 mL Et20 were stirred at room temperature for 1 day to give, after workup and silica

gel chromatog., 100 mg 1,2-bis[4-(4,4,5,5,6,6,6-heptafluoro-3-oxohexanoyl)benzyl]benzene (II). When complexed with EuCl3.6H2O,

Spr - Altra 1,2-bis[4-(4,4,5,5,6,6,6-heptafluoro-3-oxohexanoyl)phenoxy]benzene (preparation) given) exhibited the highest signal intensity in time-resolved fluorometry anti-α-fetoprotein (AFP) antibody and anti-human C reactive protein (CRP) antibody by time-resolved fluorometry was carried out using streptavidin labeled by $aryl-\beta$ -ketone-europium-complexes.

ICM C07C049-813 IC

ICS C07C049-84; C07C233-61; C07C225-22; C07C323-22; C07D233-60; G01N033-533

CC 25-16 (Benzene, Its Derivatives, and Condensed Benzenoid Compounds)

Section cross-reference(s): 9, 78

75-36-5, Acetyl chloride 91-13-4, 1,2-Bis(bromomethyl)benzene IT 101-81-5, Diphenylmethane 120-80-9, 1,2-Dihydroxybenzene, reactions 403-42-9, 4'-Fluoroacetophenone 13759-92-7, Europium chloride hexahydrate 149104-90-5, 4-Acetylphenylboronic acid

RL: RCT (Reactant); RACT (Reactant or reagent)

(preparation of aryldiketones and their complexes with rare earth elements as luminescent labeling reagents for protein, peptide, amino acid, and nucleic acid)

IT 149104-90-5, 4-Acetylphenylboronic acid

RL: RCT (Reactant); RACT (Reactant or reagent)

(preparation of aryldiketones and their complexes with rare earth elements as luminescent labeling reagents for protein, peptide, amino acid, and nucleic acid)

149104-90-5 HCAPLUS RM

Boronic acid, (4-acetylphenyl) - (9CI) (CA INDEX NAME) CN

REFERENCE COUNT:

THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

HCAPLUS COPYRIGHT 2004 ACS on STN L28 ANSWER 8 OF 36

ACCESSION NUMBER:

2003:98493 HCAPLUS

DOCUMENT NUMBER:

139:159426

TITLE:

Synthesis, structure-activity relationship and in vitro evaluation of coelenterazine and coelenteramine

derivatives as inhibitors of lipid peroxidation

Burton, Maggi; De Tollenaere, Catherine; Cavalier, Jean-Francois; Marchand, Cecile; Dussart, Frederique;

Marchand-Brynaert, Jacqueline; Rees, Jean-Francois Laboratory of Cell Biology, Institut des Sciences de

CORPORATE SOURCE: la Vie, Universite Catholique de Louvain,

Louvain-la-Neuve, B-1348, Belg.

SOURCE:

Free Radical Research (2003), 37(2), 145-158

CODEN: FRARER; ISSN: 1071-5762

PUBLISHER:

AUTHOR(S):

Taylor & Francis Ltd.

DOCUMENT TYPE:

Journal

LANGUAGE:

English

OTHER SOURCE(S):

CASREACT 139:159426

Coelenterazine (2-p-hydroxybenzyl-6-(3'-hydroxyphenyl)-8-benzyl-3,7-AB dihydroimidazolo[1,2-a]pyrazin-3-one, CLZn) and coelenteramine (2-amino-3-benzyl-5-(4'-hydroxyphenyl)-1,4-pyrazine CLM), first described as luciferin and etioluciferin, resp., of bioluminescent systems in marine organisms are endowed with antioxidant properties. was aimed at understanding the structural basis of their chain-breaking properties and at designing new compds. with improved antioxidative properties. For this, a series of 2-amino-1,4-pyrazine derivs. and their related imidazolopyrazinones were synthesized and examined for their capacity to inhibit lipid peroxidn. in linoleate micelles subjected to the peroxidizing action of AAPH. Structure-activity relation studies indicated that the reduction of the peroxidn. rate by CLM is mainly determined

by

the concomitant presence of 5-p-hydroxyphenyl and 2-amino groups in para position. The lipophilic character of substituents also affected this effect. All imidazolopyrazinones induced a lag-time before the onset of the peroxidn. process. The hetero-bicyclic imidazolopyrazinone moiety appears as the main contributor to this activity while phenol groups play little role in it. Phenol groups were required for the reduction of the peroxidn. rate after the lag-phase. The introduction of a supplementary p-hydroxyphenyl substituent at C8 position did not increase chain-breaking properties. The substitution of the C5-p-hydroxyphenyl with a catechol moiety or the introduction of a second amino group on the pyrazine ring yielded the most active compds., superior to imidazolopyrazinones and reference antioxidants like epigallocatechin gallate, vitamin E and trolox. The strong antioxidant properties of 2,6-diaminopyrazines are not dependent on the presence of hydroxyl groups indicating that their reaction mechanism differs from that of 2-amino-1,4-pyrazine derivs.

CC1-3 (Pharmacology)

Section cross-reference(s): 28

78-98-8, Methylglyoxal 98-80-6, Phenyl boronic acid 100-39-0, \mathbf{IT} Benzyl bromide 4774-14-5 5720-06-9, 2-Methoxyphenyl boronic acid 5720-07-0, 4-Methoxyphenyl boronic acid 10365-98-7 , 3-Methoxyphenyl boronic acid 14508-49-7 59489-71-3 67602-05-5 350819-24-8

RL: RCT (Reactant); RACT (Reactant or reagent)

(synthesis, structure-activity relationship of coelenterazine and coelenteramine derivs. as inhibitors of lipid peroxidn.)

98-80-6, Phenyl boronic acid 5720-06-9, 2-Methoxyphenyl boronic acid 5720-07-0, 4-Methoxyphenyl boronic acid 10365-98-7, 3-Methoxyphenyl boronic acid

RL: RCT (Reactant); RACT (Reactant or reagent)

(synthesis, structure-activity relationship of coelenterazine and coelenteramine derivs. as inhibitors of lipid peroxidn.)

98-80-6 HCAPLUS RN

Boronic acid, phenyl- (9CI) (CA INDEX NAME)

```
Ph
HO-B-OH
```

 \mathbf{IT}

CN

5720-06-9 HCAPLUS RNCN

Boronic acid, (2-methoxyphenyl) - (9CI) (CA INDEX NAME)

RN 5720-07-0 HCAPLUS

CN Boronic acid, (4-methoxyphenyl) - (9CI) (CA INDEX NAME)

RN 10365-98-7 HCAPLUS

CN Boronic acid, (3-methoxyphenyl) - (9CI) (CA INDEX NAME)

REFERENCE COUNT:

THERE ARE 21 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 9 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2001:843411 HCAPLUS

DOCUMENT NUMBER:

136:163456

TITLE:

Effect of solvents and polymers on the boronic acid

enhanced peroxidase-luminol-peroxide reaction

AUTHOR (S):

Maglia, Giovanni; Kricka, Larry J.

CORPORATE SOURCE:

Department of Pathology and Laboratory Medicine,

University of Pennsylvania Medical Center,

Philadelphia, PA, 19104, USA

SOURCE:

Bioluminescence & Chemiluminescence, Proceedings of the International Symposium, 11th, Pacific Grove, CA, United States, Sept. 6-10, 2000 (2001), Meeting Date 2000, 227-230. Editor(s): Case, James F. World Scientific Publishing Co. Pte. Ltd.: Singapore,

Singapore. CODEN: 69CAFI Conference

DOCUMENT TYPE: LANGUAGE:

Conferen English

AB The effect of both low mol. weight (MW) solvents and high MW polymer mols. (including non-hydroxy-polymers) on the 4-bromophenyl boronic acid (PBBA) enhanced chemiluminescent luminol-horseradish peroxidase (HRP)

reaction was studied. All solvents decreased the light emission from the luminol-PBBA-HRP reaction, but some solvents (e.g., dioxane) eliminated light emission even at very low concentration (8%). However, all of the solvents

tested altered the kinetics of light emission by slowing down the normal rate of light emission. Some of the polymers tested both increased and stabilized the light emission from the PBBA enhanced HRP catalyzed luminol-hydrogen peroxide reaction.

CC 9-2 (Biochemical Methods)

396097-89-5, Avitex R

Section cross-reference(s): 7

IT 64-17-5, Ethanol, analysis 67-56-1, Methanol, analysis 67-68-5, DMSO, analysis 75-12-7, Formamide, analysis 78-92-2, sec-Butanol 107-21-1, Ethylene glycol, analysis 123-51-3, Isoamyl alcohol 123-91-1, Dioxane, analysis 124-68-5, 2-Amino 2-methyl propanol 5467-74-3, 4-Bromophenyl boronic acid 25232-41-1, Poly 4-vinylpyridine 29382-68-1, Polyvinyl hydrogen phthalate 53230-15-2, Polyvinyl phthalate 54692-47-6, Zelec DP 106392-12-5, Pluronic F-68 177772-70-2, Betz 2666

RL: ARU (Analytical role, unclassified); ANST (Analytical study) (effect of solvents and polymers on boronic acid enhanced peroxidase-luminol-peroxide reaction)

IT 5467-74-3, 4-Bromophenyl boronic acid

RL: ARU (Analytical role, unclassified); ANST (Analytical study) (effect of solvents and polymers on boronic acid enhanced peroxidase-luminol-peroxide reaction)

RN 5467-74-3 HCAPLUS

CN Boronic acid, (4-bromophenyl) - (9CI) (CA INDEX NAME)

REFERENCE COUNT:

THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 10 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2001:831916 HCAPLUS

DOCUMENT NUMBER:

136:146979

TITLE:

Chemical studies on the chiral indanone derivatives as

the inhibitor of Renilla luciferase

AUTHOR(S):

Wu, Chun; Nakamura, Hideshi; Murai, Akio; Inouye,

Satoshi

CORPORATE SOURCE:

Division of Biomodeling, Department of Applied

Molecular Biosciences, Graduate School of

Bioagricultural Sciences, Nagoya University, Nagoya,

464-8601, Japan

SOURCE:

Tetrahedron (2001), 57(47), 9575-9583

CODEN: TETRAB; ISSN: 0040-4020

PUBLISHER:

Elsevier Science Ltd.

DOCUMENT TYPE:

Journal

LANGUAGE:

English

AB The bioluminescence reaction of coelenterazine involves an

oxidative process. To investigate the reaction mechanism, the authors synthesized three mechanism-based inhibitors with an indanone core structure. The inhibitors exhibited the competitive inhibition of the Renilla luciferase reaction. The (-)-4-benzyl-2-(4-hydroxybenzyl)-2hydroxymethyl-6-(4-hydroxyphenyl)-indan-1-one showed the significant enantio-selectivity of the inhibition and its absolute configuration was assigned as the R-configuration. These inhibitors could be useful probes to study the catalytic environment in the coelenterazine-luciferase reaction.

CC7-3 (Enzymes)

Section cross-reference(s): 25

4755-50-4, 4-Dimethylaminobenzoyl chloride 5720-07-0, IT 5892-99-9, 4-Bromo-N, N-diethylbenzamide 4-Methoxyphenylboronic acid 13623-25-1, 6-Methoxy-indan-1-one 15097-38-8, Benzyl (triphenylphosphoranylidene) acetate

RL: RCT (Reactant); RACT (Reactant or reagent) (reactant; preparation of chiral indanone derivs. and inhibition of Renilla luciferase)

5720-07-0, 4-Methoxyphenylboronic acid IT

RL: RCT (Reactant); RACT (Reactant or reagent) (reactant; preparation of chiral indanone derivs. and inhibition of Renilla luciferase)

RN5720-07-0 HCAPLUS

Boronic acid, (4-methoxyphenyl) - (9CI) (CA INDEX NAME) CN

REFERENCE COUNT:

20 THERE ARE 20 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 11 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2001:828964 HCAPLUS

DOCUMENT NUMBER:

135:355008

TITLE:

Determination of glycated hemoglobin by one-read method using boronate affinity capture and labeled

anti-hemoglobin antibody

INVENTOR(S):

Lee, Evelyn Mok; Westerberg, David A.; Yao, Haiou H.;

Adamczyk, Janina; Christensen, Melissa A.

PATENT ASSIGNEE(S):

Abbott Laboratories, USA

SOURCE:

U.S., 25 pp., Cont.-in-part of U.S. 6,162,645.

CODEN: USXXAM

DOCUMENT TYPE:

Patent English

LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO	KIND	DATE	APPLICATION NO.	DATE
US 6316265	B1	20011113	US 1999-399219	19990917
US 6162645	Α	20001219	US 1997-816237	19970313
ES 2206900	Т3	20040516	ES 1998-910389	19980313

```
WO 2000-US24159
     WO 2001020338
                          A1
                                20010322
         W: CA, JP
        RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
             PT, SE
                                20020612
                                            EP 2000-964939
                          A1
     EP 1212623
         R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, FI, CY
     JP 2003509693
                          Т2
                                20030311
                                            JP 2001-523873
                                                                   20000905
                                            US 1997-816237
                                                                A2 19970313
PRIORITY APPLN. INFO.:
                                            US 1999-399219
                                                                A 19990917
                                                                W 20000905
                                            WO 2000-US24159
     A glycated Hb assay utilizes a simple procedure for the determination of
AB
     standardized GHb in whole blood samples correlated to the Diabetes Control
     and Complications Trial (DCCT). First, a lysed whole blood sample is
     incubated with a solid phase that is coupled with boronic acid or similar
     boronate compound through covalent linkage chemistries known in the art.
     Next, a labeled antibody to human Hb is added and the resulting signal is
     directly proportional to the GHb in the sample. The advantages of
     measuring GHb using a single determination include high precision and, since
the
     assay is easily automatable, high throughput. With automation, this assay
     can also be consolidated with other testing on one analyzer. The method
     according to the various embodiments of the invention thus eliminates the
     need for two measurements: one for GHb and another for total Hb (THb).
     ICM G01N033-72
     436067000
NCL
     9-10 (Biochemical Methods)
CC
     Chemiluminescent substances
IT
     Fluorescent substances
     Radioactive substances
        (as labels; determination of glycated Hb by one-read method using boronate
        affinity capture and labeled anti-Hb antibody)
     30418-59-8DP, m-Aminophenylboronic acid, immobilized
IT
     RL: ARG (Analytical reagent use); DEV (Device component use); SPN
     (Synthetic preparation); THU (Therapeutic use); ANST (Analytical study);
     BIOL (Biological study); PREP (Preparation); USES (Uses)
        (affinity complex; determination of glycated Hb by one-read method using
        boronate affinity capture and labeled anti-Hb antibody)
     98-80-6D, Phenylboronic acid, compds., immobilized
                                                          10043-35-3D,
IT
                               13780-71-7D, Boronic acid, compds., immobilized
     Boric acid, immobilized
     14047-29-1D, 4-Carboxyphenylboronic acid, immobilized
     101084-81-5D, immobilized
     RL: ARG (Analytical reagent use); DEV (Device component use); THU
     (Therapeutic use); ANST (Analytical study); BIOL (Biological study); USES
     (Uses)
        (affinity complex; determination of glycated Hb by one-read method using
        boronate affinity capture and labeled anti-Hb antibody)
     107-15-3, Ethylenediamine, reactions
IT
                                           4097-89-6, Tris(2-aminoethyl)amine
     9003-01-4, Polyacrylic acid
                                   9004-32-4, Carboxymethylcellulose
     12768-31-9, Carboxymethylamylose 66472-86-4,
     m-Aminophenylboronic acid (hemisulfate)
                                               330582-39-3, AM 40-500
     330582-42-8, SP1267
                          330582-49-5, SP1340
                                               330582-60-0, AB007C
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (determination of glycated Hb by one-read method using boronate affinity
        capture and labeled anti-Hb antibody)
     30418-59-8DP, m-Aminophenylboronic acid, immobilized
IT
     RL: ARG (Analytical reagent use); DEV (Device component use); SPN
     (Synthetic preparation); THU (Therapeutic use); ANST (Analytical study);
     BIOL (Biological study); PREP (Preparation); USES (Uses)
```

(affinity complex; determination of glycated Hb by one-read method using boronate affinity capture and labeled anti-Hb antibody)

RN 30418-59-8 HCAPLUS

CN Boronic acid, (3-aminophenyl) - (9CI) (CA INDEX NAME)

98-80-6D, Phenylboronic acid, compds., immobilized 14047-29-1D, 4-Carboxyphenylboronic acid, immobilized

101084-81-5D, immobilized

RL: ARG (Analytical reagent use); DEV (Device component use); THU (Therapeutic use); ANST (Analytical study); BIOL (Biological study); USES (Uses)

(affinity complex; determination of glycated Hb by one-read method using boronate affinity capture and labeled anti-Hb antibody)

RN 98-80-6 HCAPLUS

CN Boronic acid, phenyl- (9CI) (CA INDEX NAME)

RN 14047-29-1 HCAPLUS CN Benzoic acid, 4-borono- (9CI) (CA INDEX NAME)

RN 101084-81-5 HCAPLUS CN Benzoic acid, 3-borono-5-nitro- (6CI, 9CI) (CA INDEX NAME)

66472-86-4, m-Aminophenylboronic acid (hemisulfate) IT

RL: RCT (Reactant); RACT (Reactant or reagent)

(determination of glycated Hb by one-read method using boronate affinity capture and labeled anti-Hb antibody)

66472-86-4 HCAPLUS RN

Boronic acid, (3-aminophenyl)-, sulfate (2:1) (9CI) (CA INDEX NAME) CN

CM

CRN 30418-59-8 CMF C6 H8 B N O2

CM

7664-93-9 CRN H2 O4 S CMF

REFERENCE COUNT:

THERE ARE 32 CITED REFERENCES AVAILABLE FOR THIS 32 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 12 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2001:382816 HCAPLUS

DOCUMENT NUMBER:

135:88932

TITLE:

Application of an enhanced luminol chemiluminescence reaction using

4-[4,5-di(2-pyridyl)-1H-imidazol-2-yl]phenylboronic

acid to photographic detection of horseradish

peroxidase on a membrane

AUTHOR(S):

SOURCE:

Kuroda, Naotaka; Murasaki, Naoko; Wada, Mitsuhiro;

Nakashima, Kenichiro

CORPORATE SOURCE:

School of Pharmaceutical Sciences, Nagasaki

University, Nagasaki, 852-8521, Japan

Luminescence (2001), 16(2), 167-172 CODEN: LUMIFC; ISSN: 1522-7235

John Wiley & Sons Ltd.

PUBLISHER: DOCUMENT TYPE:

Journal

English

LANGUAGE:

Photog. detection of horseradish peroxidase (HRP) on a membrane by the

luminol-H2O2-HRP chemiluminescence reaction using 4-[4,5-di(2-pyridyl)-1H-imidazol-2-yl]phenylboronic acid (DPPA) as an enhancer is described. The method is based on the long-lived chemiluminescence emission obtained by using DPPA. Under the optimum conditions, as little as 0.10 ng (.apprx.2.3 fmol) and 0.20 ng (.apprx.4.6 fmol) per spot of HRP on a membrane were detected as visible spots with exposure times of 60 and 10 min, resp., by using an instant photog. film and a camera luminometer. The proposed method was highly sensitive and was successfully applied to the detection of HRP conjugates as an alternative to the colorimetric method using a chromogenic substrate in a com. available assay kit of Western blotting.

CC 7-1 (Enzymes)

ST peroxidase detn membrane chemiluminescence photog detection

IT Chemiluminescence spectroscopy

(application of enhanced luminol chemiluminescence reaction using a phenylboronic acid derivative to photog. detection of horseradish peroxidase and its conjugates on a membrane)

IT Membranes, nonbiological

(nitrocellulose; application of enhanced luminol chemiluminescence reaction using a phenylboronic acid derivative to photog. detection of horseradish peroxidase and its conjugates on a membrane)

IT 9003-99-0D, Peroxidase, horseradish, conjugates

RL: ANT (Analyte); ANST (Analytical study)
(application of enhanced luminol chemiluminescence reaction
using a phenylboronic acid derivative to photog. detection of horseradish
peroxidase and its conjugates on a membrane)

IT 521-31-3, Luminol 7722-84-1, Hydrogen peroxide, uses
RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
(application of enhanced luminol chemiluminescence reaction
using a phenylboronic acid derivative to photog. detection of horseradish
peroxidase and its conjugates on a membrane)

IT 264889-15-8

RL: ARU (Analytical role, unclassified); ANST (Analytical study) (application of enhanced luminol chemiluminescence reaction using a phenylboronic acid derivative to photog. detection of horseradish peroxidase and its conjugates on a membrane)

IT 9003-99-0, Peroxidase

RL: ANT (Analyte); ANST (Analytical study)
(horseradish; application of enhanced luminol chemiluminescence
reaction using a phenylboronic acid derivative to photog. detection of
horseradish peroxidase and its conjugates on a membrane)

IT 264889-15-8

RL: ARU (Analytical role, unclassified); ANST (Analytical study) (application of enhanced luminol chemiluminescence reaction using a phenylboronic acid derivative to photog. detection of horseradish peroxidase and its conjugates on a membrane)

RN 264889-15-8 HCAPLUS

CN Boronic acid, [4-(4,5-di-2-pyridinyl-1H-imidazol-2-yl)phenyl]- (9CI) (CA INDEX NAME)

REFERENCE COUNT:

THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS 12 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 13 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2001:208515 HCAPLUS

DOCUMENT NUMBER:

134:234022

TITLE:

Determination of % glycated hemoglobin using

immobilized boronate and labeled antibody

INVENTOR(S):

Lee, Evelyn Mok; Westerberg, David A.; Yao, Haiou H.;

Adamczyk, Janina; Christensen, Melissa A.

PATENT ASSIGNEE(S):

SOURCE:

Abbott Laboratories, USA

PCT Int. Appl., 38 pp.

CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

WO 2001020338 A1 20010322 WO 2000-US24159 200	000905
W: CA, JP	
RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, I	MC, NL,
•	990917
Br 1212025 M1 20020012 21 2000 701707 = 1	000905
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE,	MC, PT,
IE, FI, CY TP 2003509693 T2 20030311 JP 2001-523873 20	000905
0F 2003307073 12 20030311 01 2001 020070 =-	990917
PRIORITI ALLIM: INIO::	970313
WO 2000-US24159 W 200	000905

A glycated Hb (GHb) assay utilizes a simple procedure for the determination of AB diabetes control and complications trial (DCCT) standardized %GHb in whole blood samples. First, a lysed whole blood sample is incubated with a solid phase that is coupled with boronic acid or similar boronate compound through covalent linkage chemistries known in the art. Next, a labeled antibody to human Hb is added and the resulting signal is directly proportional to the %GHb in the sample. The advantages of measuring %GHb using a single determination include high precision and, since the assay is

easily

automatable, high throughput. With automation, this assay can also be consolidated with other testing on one analyzer. The method according to the various embodiments of the invention thus eliminates the need for two measurements: one for GHb and another for total Hb (THb).

```
ICM G01N033-72
IC
     ICS G01N033-543
     9-10 (Biochemical Methods)
CC
     Chemiluminescent substances
IT
     Fluorescent substances
     Radioactive substances
        (as labels; determination of % glycated Hb using immobilized boronate and
        labeled antibody)
     Immunoassay
IT
        (chemiluminescence; determination of % glycated Hb using immobilized
        boronate and labeled antibody)
     98-80-6D, Phenylboronic acid, immobilized
                                                 10043-35-3D, Boric
IT
                        13780-71-7D, Boronic acid, immobilized
     acid, immobilized
     14047-29-1D, 4-Carboxyphenylboronic acid, immobilized
     30418-59-8D, m-Aminophenylboronic acid, immobilized
     101084-81-5D, immobilized
     RL: ARG (Analytical reagent use); BPR (Biological process); BSU
     (Biological study, unclassified); THU (Therapeutic use); ANST (Analytical
     study); BIOL (Biological study); PROC (Process); USES (Uses)
        (determination of % glycated Hb using immobilized boronate and labeled
        antibody)
                                            4097-89-6, Tris(2-aminoethyl)amine
     107-15-3, Ethylenediamine, reactions
IT
     66472-86-4, m-Aminophenylboronic acid hemisulfate
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (determination of % glycated Hb using immobilized boronate and labeled
        antibody)
     98-80-6D, Phenylboronic acid, immobilized 14047-29-1D,
TT
     4-Carboxyphenylboronic acid, immobilized 30418-59-8D,
     m-Aminophenylboronic acid, immobilized 101084-81-5D, immobilized
     RL: ARG (Analytical reagent use); BPR (Biological process); BSU
     (Biological study, unclassified); THU (Therapeutic use); ANST (Analytical
     study); BIOL (Biological study); PROC (Process); USES (Uses)
        (determination of % glycated Hb using immobilized boronate and labeled
        antibody)
     98-80-6 HCAPLUS
RN
     Boronic acid, phenyl- (9CI) (CA INDEX NAME)
CN
   Ph
HO- B- OH
```

RN 14047-29-1 HCAPLUS CN Benzoic acid, 4-borono- (9CI) (CA INDEX NAME)

RN 30418-59-8 HCAPLUS

المباري والمراب وأكربي والموقوعيات المراويها والأألية

CN Boronic acid, (3-aminophenyl) - (9CI) (CA INDEX NAME)

RN 101084-81-5 HCAPLUS CN Benzoic acid, 3-borono-5-nitro- (6CI, 9CI) (CA INDEX NAME)

IT 66472-86-4, m-Aminophenylboronic acid hemisulfate
RL: RCT (Reactant); RACT (Reactant or reagent)
 (determination of % glycated Hb using immobilized boronate and labeled antibody)

RN 66472-86-4 HCAPLUS

CN Boronic acid, (3-aminophenyl)-, sulfate (2:1) (9CI) (CA INDEX NAME)

CM 1

CRN 30418-59-8 CMF C6 H8 B N O2

CM 2

CRN 7664-93-9 CMF H2 O4 S

10

```
OH
```

REFERENCE COUNT:

THERE ARE 1 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 14 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

1

ACCESSION NUMBER:

2001:169756 HCAPLUS

DOCUMENT NUMBER:

134:353639

TITLE:

Polymeric alkoxy PBD [2-(4-biphenylyl)-5-phenyl-1,3,4-

oxadiazole] for light-emitting diodes

AUTHOR (S):

Wang, Changsheng; Kilitziraki, Mary; Palsson, Lars-Olof; Bryce, Martin R.; Monkman, Andrew P.;

Samuel, Ifor D. W.

CORPORATE SOURCE:

Department of Chemistry, University of Durham, Durham,

DH1 3LE, UK

SOURCE:

Advanced Functional Materials (2001), 11(1), 47-50

CODEN: AFMDC6; ISSN: 1616-301X

PUBLISHER:

Wiley-VCH Verlag GmbH

DOCUMENT TYPE:

Journal English

LANGUAGE:

Polymeric alkoxy [2-(4-biphenylyl)-5-phenyl-1,3,4-oxadiazole] derivs. (I & II) were prepared using Suzuki coupling reactions of 1,4-dialkoxybenzene-2,5diboronic acid with 2,5-bis(4-bromophenyl)-1,3,5-oxadiazole, and its dipyridyl analog, resp. Thermal gravimetric anal. shows that the polymers are stable up to 370° and 334°, resp. Films of polymer I spun from chloroform solution show an absorption at $\lambda max = 367$ nm, and a weaker band at 312 nm, and strong blue photoluminescence at λmax = 444 nm. The photoluminescence quantum yield (PLQY) was found to be 27 \pm 3%. For polymer II, the absorption spectra reveal bands of equal intensity at $\lambda max = 374$ and 312 nm, with PL at $\lambda max = 475$ nm. Device studies using polymer II were hampered by its instability under illumination and/or elec. excitation. Polymer I is stable under these conditions and acts as an efficient electron-transporting/holeblocking layer. For devices of configuration ITO/PEDOT/MEH-PPV/polymer I/Al an external quantum efficiency of 0.26% and brightness of 800 cd/m2 was readily achieved: orange emission was observed, identical to the MEH-PPV electroluminescence.

35-7 (Chemistry of Synthetic High Polymers) CC

Section cross-reference(s): 73

Luminescence, chemiluminescence IT

Luminescence, electroluminescence

(in preparation of polymeric alkoxy PBD [2-(4-biphenylyl)-5-phenyl-1,3,4oxadiazole])

106-51-4, 2,5-Cyclohexadiene-1,4-dione, reactions 115-19-5, TT2-Methyl-3-butyn-2-ol 586-75-4, 4-Bromobenzoyl chloride 624-28-2, 2,5-Dibromopyridine 5933-32-4, p-Bromobenzoic hydrazide 10035-10-6, Hydrobromic acid, reactions 14753-51-6, 2,5-Dibromohydroquinone 19542-05-3 69673-99-0 **123324-71-0**, 4-tert-Butylphenylboronic

RL: RCT (Reactant); RACT (Reactant or reagent)

(in preparation of polymeric alkoxy PBD [2-(4-biphenyly1)-5-phenyl-1,3,4oxadiazole])

30766-11-1P, 5-Bromopyridine, 2-carboxylic acid 134321-95-2P IT

191867-85-3P 191917-63-2P 339064-91-4P 339064-93-6P RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (in preparation of polymeric alkoxy PBD [2-(4-biphenyly1)-5-phenyl-1,3,4oxadiazole]) 339065-01-9P IT 339064-96-9P RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (photoluminescence; in preparation of polymeric alkoxy PBD [2-(4-biphenylyl)-5-phenyl-1,3,4-oxadiazole]) IT 339064-88-9P RL: DEV (Device component use); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses) (photoluminescence; preparation of polymeric alkoxy PBD [2-(4-biphenyly1)-5phenyl-1,3,4-oxadiazole] for light-emitting diodes) 123324-71-0, 4-tert-Butylphenylboronic acid IT RL: RCT (Reactant); RACT (Reactant or reagent) (in preparation of polymeric alkoxy PBD [2-(4-biphenyly1)-5-phenyl-1,3,4oxadiazole]) RN123324-71-0 HCAPLUS

Boronic acid, [4-(1,1-dimethylethyl)phenyl]- (9CI) (CA INDEX NAME)

CN

IT

RN

CN

IT 191917-63-2P
 RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
 (Reactant or reagent)
 (in preparation of polymeric alkoxy PBD [2-(4-biphenylyl)-5-phenyl-1,3,4-oxadiazole])
RN 191917-63-2 HCAPLUS
CN Boronic acid, [2,5-bis[(2-ethylhexyl)oxy]-1,4-phenylene]bis- (9CI) (CAINDEX NAME)

$$\begin{array}{c|c} \text{OH} & \text{Et} \\ & \text{HO-B} & \text{O-CH}_2\text{-CH-Bu-n} \\ \\ \text{Et} & \\ & \text{n-Bu-CH-CH}_2\text{-O} & \\ & \text{B-OH} \\ & \text{OH} \end{array}$$

339064-96-9P
RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
 (photoluminescence; in preparation of polymeric alkoxy PBD
 [2-(4-biphenyly1)-5-phenyl-1,3,4-oxadiazole])
339064-96-9 HCAPLUS
Boronic acid, [2,5-bis[(2-ethylhexyl)oxy]-1,4-phenylene]bis-, polymer with

2,2'-(1,3,4-oxadiazole-2,5-diyl)bis[5-bromopyridine] (9CI) (CA INDEX NAME)

CM 1

CRN 339064-93-6 CMF C12 H6 Br2 N4 O

CM 2

CRN 191917-63-2 CMF C22 H40 B2 O6

$$\begin{array}{c|c} \text{OH} & \text{Et} \\ & & \\ \text{HO-B} & \text{O-CH}_2\text{-CH-Bu-n} \\ \\ \text{Et} & \\ \text{n-Bu-CH-CH}_2\text{-O} & \\ & & \\ \text{B-OH} \\ & \\ \text{OH} \end{array}$$

IT 339064-88-9P

RL: DEV (Device component use); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)

(photoluminescence; preparation of polymeric alkoxy PBD [2-(4-biphenyly1)-5-phenyl-1,3,4-oxadiazole] for light-emitting diodes)

RN 339064-88-9 HCAPLUS

CN Boronic acid, [2,5-bis[(2-ethylhexyl)oxy]-1,4-phenylene]bis-, polymer with 2,5-bis(4-bromophenyl)-1,3,4-oxadiazole (9CI) (CA INDEX NAME)

CM 1

CRN 191917-63-2 CMF C22 H40 B2 O6

$$\begin{array}{c|c} \text{OH} & \text{Et} \\ \text{HO-B} & \text{O-CH}_2\text{-CH-Bu-n} \\ \\ \text{N-Bu-CH-CH}_2\text{-O} & \text{B-OH} \\ \\ \text{OH} & \\ \end{array}$$

CM 2

CRN 19542-05-3 CMF C14 H8 Br2 N2 O

REFERENCE COUNT:

THERE ARE 32 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 15 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2001:58741 HCAPLUS

DOCUMENT NUMBER:

134:262751

TITLE:

Design, Synthesis, and Evaluation of the Transition-

State Inhibitors of Coelenterazine Bioluminescence: Probing the Chiral

Environment of Active Site

AUTHOR (S):

Nakamura, Hideshi; Wu, Chun; Inouye, Satoshi; Murai,

Akio

CORPORATE SOURCE:

Division of Biomodeling Department of Applied

Molecular Biosciences Graduate School of

Bioagricultural Sciences, Nagoya University, Nagoya,

464-8601, Japan

SOURCE:

Journal of the American Chemical Society (2001),

123(7), 1523-1524

CODEN: JACSAT; ISSN: 0002-7863

American Chemical Society

PUBLISHER:
DOCUMENT TYPE:

Journal

LANGUAGE:

English

OTHER SOURCE(S):

CASREACT 134:262751

AB The authors have established a novel and effective route to synthesize stable, chiral analogs of coelenterazine for studying transition states and unstable intermediates involved in the **bioluminescence** of coelenterazine. Among the analogs studied, the hydroxymethyl model of the hydroperoxide structure, (R)-TS-2, showed the most potent inhibition of

recombinant Renilla luciferase. Efforts to elucidate the key structure with the required for efficient luminescence and the identification of catalytic the required for efficient luminescence and the identification of catalytic the required for efficient luminescence and the identification of catalytic the required for efficient luminescence and the identification of catalytic the required for efficient luminescence and the identification of catalytic the required for efficient luminescence and the identification of catalytic the required for efficient luminescence and the identification of catalytic the required for efficient luminescence and the identification of catalytic the required for efficient luminescence and the identification of catalytic the required for efficient luminescence and the identification of catalytic the required for efficient luminescence and the identification of catalytic the required for efficient luminescence and the identification of catalytic the required for efficient luminescence and the identification of catalytic the required for efficient luminescence and the identification of catalytic the required for efficient luminescence and the required for efficient The American State of the American Actions of the Amer site are in progress.

CC 7-4 (Enzymes)

Section cross-reference(s): 26

luciferase transition state inhibitor prepn coelenterazine STbioluminescence

Luminescence, bioluminescence IT

(design, synthesis, and evaluation of luciferase transition-state inhibitors of coelenterazine bioluminescence)

331816-24-1P 331816-25-2P 331816-35-4P 331816-36-5P IT RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation)

(design, synthesis, and evaluation of luciferase transition-state inhibitors of coelenterazine bioluminescence)

61869-41-8, Luciferase IT

RL: BPR (Biological process); BSU (Biological study, unclassified); BIOL (Biological study); PROC (Process)

(design, synthesis, and evaluation of luciferase transition-state inhibitors of coelenterazine bioluminescence)

55779-48-1, Coelenterazine IT

RL: BSU (Biological study, unclassified); BIOL (Biological study) (design, synthesis, and evaluation of luciferase transition-state inhibitors of coelenterazine bioluminescence)

IT 331816-26-3P

RL: PUR (Purification or recovery); SPN (Synthetic preparation); PREP (Preparation)

(design, synthesis, and evaluation of luciferase transition-state inhibitors of coelenterazine bioluminescence)

IT5892-99-9

RL: RCT (Reactant); RACT (Reactant or reagent) (design, synthesis, and evaluation of luciferase transition-state inhibitors of coelenterazine bioluminescence)

331816-29-6P 331816-30-9P 331816-31-0P 331816-27-4P 331816-28-5P TT331816-34-3P 331816-32-1P 331816-33-2P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(design, synthesis, and evaluation of luciferase transition-state inhibitors of coelenterazine bioluminescence)

5720-07-0, 4-Methoxyphenylboronic acid 15097-38-8, Benzyl TT

(triphenylphosphoranylidene)acetate

RL: RCT (Reactant); RACT (Reactant or reagent) (preparation of)

5720-07-0, 4-Methoxyphenylboronic acid IT

RL: RCT (Reactant); RACT (Reactant or reagent) (preparation of)

5720-07-0 HCAPLUS

Boronic acid, (4-methoxyphenyl)- (9CI) (CA INDEX NAME) CN

RN

REFERENCE COUNT:

THERE ARE 34 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 16 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2000:248612 HCAPLUS

DOCUMENT NUMBER:

133:17320

TITLE:

Convergent and short-step syntheses of dl-Cypridina luciferin and its analogs based on Pd-mediated cross

couplings

AUTHOR (S):

Nakamura, Hideshi; Aizawa, Mihoko; Takeuchi, Daisuke;

Murai, Akio; Shimoura, Osamu

CORPORATE SOURCE:

Division of Biomodeling, Department of Applied

Molecular Biosciences, Graduate School of

Bioagricultural Sciences, Nagoya University, Nagoya,

464-8601, Japan

SOURCE:

Tetrahedron Letters (2000), 41(13), 2185-2188

CODEN: TELEAY; ISSN: 0040-4039

PUBLISHER:

Elsevier Science Ltd.

DOCUMENT TYPE:

Journal English

LANGUAGE:
OTHER SOURCE(S):

CASREACT 133:17320

AB (±)-Cypridina luciferin and its analogs I (X = NH, O, S) were synthesized from 2-aminopyrazine by an eight-step method that included two regio-selective Pd-mediated cross couplings, and their chemi- and bioluminescent activities were compared. Analogs having a 3-benzofuranyl or a 3-benzothienyl group in the place of a 3-indolyl group showed luciferase affinities similar to Cypridina luciferase but with a lower luminescent efficiency, suggesting that the NH group is unimportant for mol. recognition whereas the indolyl group is crucial for efficient luminescence.

CC 26-9 (Biomolecules and Their Synthetic Analogs)

Section cross-reference(s): 7

ST Cypridina luciferin oxa thia analog prepn; Suzuki coupling palladium mediated aminobromopyrazine indole; chemiluminescence Cypridina luciferin oxa thia analog; bioluminescence Cypridina luciferin oxa thia analog

IT Heterocyclization

Luminescence, bioluminescence

Luminescence, chemiluminescence

(preparation of (\pm) -Cypridina luciferin and its oxa and thia analogs via Pd-mediated cross couplings)

IT 24241-18-7 92136-39-5 142913-28-8 **149108-61-2** 271260-62-9 271260-63-0

RL: RCT (Reactant); RACT (Reactant or reagent)

(preparation of (\pm) -Cypridina luciferin and its oxa and thia analogs via Pd-mediated cross couplings)

IT 149108-61-2

RL: RCT (Reactant); RACT (Reactant or reagent)

(preparation of (\pm) -Cypridina luciferin and its oxa and thia analogs via Pd-mediated cross couplings)

RN 149108-61-2 HCAPLUS

CN Boronic acid, [1-[(4-methylphenyl)sulfonyl]-1H-indol-3-yl]- (9CI) (CA INDEX NAME)

REFERENCE COUNT:

THERE ARE 21 CITED REFERENCES AVAILABLE FOR THIS 21 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 17 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

2000:87182 HCAPLUS

DOCUMENT NUMBER:

132:290260

TITLE:

New phenylboronic acid derivatives as enhancers of the

luminol-H2O2-horseradish peroxidase

chemiluminescence reaction

AUTHOR (S):

Kuroda, Naotaka; Kawazoe, Kaori; Nakano, Hirofumi; Wada, Mitsuhiro; Nakashima, Kenichiro

CORPORATE SOURCE:

School of Pharmaceutical Sciences, Nagasaki

University, Nagasaki, 852-8521, Japan

SOURCE:

Luminescence (1999), 14(6), 361-364 CODEN: LUMIFC; ISSN: 1522-7235

PUBLISHER:

John Wiley & Sons Ltd.

DOCUMENT TYPE:

Journal

English LANGUAGE:

The preparation of three new types of phenylboronic acid derivs. and their evaluation as enhancers on the luminol-H2O2-horseradish peroxidase (HRP) chemiluminescence (CL) reaction are described. After optimizing the CL reaction conditions, the CL system was applied to the HRP determination Among the three phenylboronic acid derivs., i.e. 4-(4,5-diphenyl-1Himidazol-2-yl)phenylboronic acid (DPA), 4-[4(or 5)-(4-dimethylaminophenyl)-5 (or 4) -phenyl-1H-imidazol-2-yl]phenylboronic acid (DAPA) and 4-[4,5-di(2-pyridyl)-1H-imidazol-2-yl]phenylboronic acid (DPPA), DPPA was found to be the most potent enhancer. The sensitivity obtained with DPPA was about 180 times higher than that without an enhancer. The detection limit of HRP obtained with DPPA was 0.15 ng/assay (ca. 3.5 fmol), which is comparable to that with 4-iodophenol under the conditions examined All the phenylboronic acid derivs. examined had the effect of prolonging light emission compared to 4-iodophenol.

CC7-1 (Enzymes)

peroxidase detn chemiluminescence enhancement phenylboronate ST deriv

Luminescence, chemiluminescence IT

(new phenylboronic acid derivs. as enhancers of the luminol-H2O2-horseradish peroxidase chemiluminescence reaction)

9003-99-0, Peroxidase IT

RL: ANT (Analyte); ANST (Analytical study)
(new phenylboronic acid derivs. as enhancers of the
luminol-H2O2-horseradish peroxidase chemiluminescence
reaction)

IT 521-31-3, Luminol

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (new phenylboronic acid derivs. as enhancers of the luminol-H2O2-horseradish peroxidase chemiluminescence reaction)

IT 264889-14-7P 264889-15-8P 264889-16-9P

RL: ARU (Analytical role, unclassified); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation)

(new phenylboronic acid derivs. as enhancers of the luminol-H2O2-horseradish peroxidase chemiluminescence reaction)

TT 7722-84-1, Hydrogen peroxide, biological studies
RL: BPR (Biological process); BSU (Biological study, unclassified); BIOL
(Biological study); PROC (Process)

(new phenylboronic acid derivs. as enhancers of the luminol-H2O2-horseradish peroxidase chemiluminescence reaction)

IT 134-81-6, Benzil 492-73-9, 2,2'-Pyridil 22711-20-2, 4-Dimethylaminobenzil 87199-17-5, 4-Formylphenylboronic acid RL: RCT (Reactant); RACT (Reactant or reagent) (new phenylboronic acid derivs. as enhancers of the luminol-H202-borseradish peroxidase chemiluminescence

luminol-H2O2-horseradish peroxidase chemiluminescence reaction)

264889-14-7P 264889-15-8P 264889-16-9P

IT 264889-14-7P 264889-15-8P 264889-16-9P RL: ARU (Analytical role, unclassified); SPN (Synthetic preparation); ANST

(Analytical study); PREP (Preparation)
 (new phenylboronic acid derivs. as enhancers of the
 luminol-H2O2-horseradish peroxidase chemiluminescence
 reaction)

RN 264889-14-7 HCAPLUS

CN Boronic acid, [4-(4,5-diphenyl-1H-imidazol-2-yl)phenyl]- (9CI) (CA INDEX NAME)

RN 264889-15-8 HCAPLUS

CN Boronic acid, [4-(4,5-di-2-pyridinyl-1H-imidazol-2-yl)phenyl]- (9CI) (CF INDEX NAME)

RN 264889-16-9 HCAPLUS

CN Boronic acid, [4-[4-[4-(dimethylamino)phenyl]-5-phenyl-1H-imidazol-2-yl]phenyl]- (9CI) (CA INDEX NAME)

IT 87199-17-5, 4-Formylphenylboronic acid

RL: RCT (Reactant); RACT (Reactant or reagent)
(new phenylboronic acid derivs. as enhancers of the
luminol-H2O2-horseradish peroxidase chemiluminescence
reaction)

RN 87199-17-5 HCAPLUS

CN Boronic acid, (4-formylphenyl) - (9CI) (CA INDEX NAME)

REFERENCE COUNT:

THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 18 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1998:795186 HCAPLUS

DOCUMENT NUMBER:

130:35358

TITLE: INVENTOR(S): Chemiluminescent hemoglobin assay

Hixson, Craig S.

PATENT ASSIGNEE(S):

Bio-Rad Laboratories, Inc., USA

SOURCE:

PCT Int. Appl., 17 pp.

CODEN: PIXXD2

```
DOCUMENT TYPE:
```

Patent English

LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

```
KIND
                               DATE
                                          APPLICATION NO.
     PATENT NO.
                        ----
     _____
                               ------
     WO 9854578
                               19981203 WO 1998-US9867
                        A1
        W: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE,
            DK, EE, ES, FI, GB, GE, GH, GM, GW, HU, ID, IL, IS, JP, KE, KG,
            KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX,
            NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT,
            UA, UG, US, UZ, VN, YU, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
        RW: GH, GM, KE, LS, MW, SD, SZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES,
            FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI,
             CM, GA, GN, ML, MR, NE, SN, TD, TG
                               19981230
                                          AU 1998-74877
                                                                  19980514
     AU 9874877
                         A1
PRIORITY APPLN. INFO.:
                                           US 1997-865367
                                                                  19970529
                                           WO 1998-US9867
                                                                  19980514
     The Hb content of a sample is determined by chemiluminescence, based
AB
     on the ability of Hb to absorb radiation emitted by a
     chemiluminescent reaction. To perform the assay, the sample is
     placed in a liquid medium with a compound that is susceptible to a
     chemiluminescent reaction, and the medium is exposed to conditions
     causing the chemiluminescent reaction to occur. The amount of
     emission created by the reaction and not absorbed by the Hb is then
     detected and compared to calibrators or stds. as a measure of the amount of
     Hb present. By combining the above with procedures that sep. and quantify
     glycosylated Hb, the assay can be used to determine glycosylated Hb as a
     proportion of the total.
IC
     ICM G01N033-72
     ICS G01N033-543; G01N033-551
     9-5 (Biochemical Methods)
CC
     chemiluminescent Hb assay
ST
     Onium compounds
IT
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (acridinium; chemiluminescent Hb assay)
     Chemiluminescence spectroscopy
IT
     Particles
        (chemiluminescent Hb assay)
IT
     Hemoglobins
     RL: ANT (Analyte); ANST (Analytical study)
        (chemiluminescent Hb assay)
     Alkali metal hydroxides
IT
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (chemiluminescent Hb assay)
IT
     Hemoglobins
     RL: ANT (Analyte); ANST (Analytical study)
        (glycohemoglobins; chemiluminescent Hb assay)
IT
     Particles
        (paramagnetic; chemiluminescent Hb assay)
     98-80-6, Phenylboronic acid 144-62-7D, Ethanedioic acid,
IT
     compds., uses 7697-37-2, Nitric acid, uses 7722-84-1, Hydrogen
     peroxide (H2O2), uses 13780-71-7D, Boronic acid, alkane 14915-07-2,
               113630-26-5 216774-74-2D, compds.
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (chemiluminescent Hb assay)
     98-80-6, Phenylboronic acid
IT
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
```

(chemiluminescent Hb assay)

RN 98-80-6 HCAPLUS

CN Boronic acid, phenyl- (9CI) (CA INDEX NAME)

Ph | HO— B— OH

REFERENCE COUNT:

17 THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 19 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1998:621381 HCAPLUS

DOCUMENT NUMBER:

129:242239

TITLE:

Determination of % glycated hemoglobin

INVENTOR (S):

Lee, Evelyn Mok; Westerberg, David A.; Yao, Haiou H.;

Adamczyk, Janina; Christensen, Melissa A.

PATENT ASSIGNEE(S):

Abbott Laboratories, USA

SOURCE:

PCT Int. Appl., 45 pp.

CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

	PA'	TENT NO			KIN	D DATE	APPLICATION NO.		DATE	
	WO	984075 W: C	0		A1	1998091	7 WO 1998-US5005	-	19980313	
				CH,	DE,	DK, ES, FI	, FR, GB, GR, IE, IT,	LU, M	C, NL, PT,	SE
	US	616264			A	2000121			19970313	
	EP	974060			A1	2000012	6 EP 1998-910389		19980313	
	EP	974060			В1	2003090	3			
		R: A'	T, BE,	CH,	DE,	ES, FR, GB	, IT, LI, NL			
	JP	200051			T 2	2000102			19980313	
	JP	335678	5		B2	2002121	6			
	AΤ	249049			\mathbf{E}	2003091	5 AT 1998-910389		19980313	
	ES	220690	0		Т3	2004051	6 ES 1998-910389		19980313	
PRI	ORIT	Y APPLN	. INFO	. :			US 1997-816237	Α	19970313	
							WO 1998-US5005	W	19980313	

AB A glycated Hb assay utilizes a simple procedure for the determination of DCCT Standardized %GHb in whole blood samples. First, a lysed whole blood sample is incubated with a solid phase that is coupled with boronic acid or similar boronate compound through covalent linkage chemistries known in the art. Next, a labeled antibody to human Hb is added and the resulting signal is directly proportional to the %GHb in the sample. The advantages of measuring %GHb using a single determination include high precision and,

since

the assay is easily automatable, high throughput. With automation, this assay can also be consolidated with other testing on one analyzer. The method according to the various embodiments of the invention thus eliminates the need for two measurements: one for GHb and another for total Hb (THb).

IC ICM G01N033-72

ICS G01N033-543

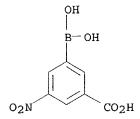
CC 9-16 (Biochemical Methods)

ITBlood analysis Chemiluminescent substances Fluorescent substances Isotope indicators Microparticles Microtiter plates Pipes and Tubes (determination of percentage glycated Hb) IT 98-80-6, Phenylboronic acid 13780-71-7, Boronic acid 13780-71-7D, Boronic acid, compds. 14047-29-1, 4-Carboxyphenylboronic acid 30418-59-8, m-Aminophenylboronic acid 101084-81-5 RL: ARU (Analytical role, unclassified); ANST (Analytical study) (determination of percentage glycated Hb) IT 98-80-6, Phenylboronic acid 14047-29-1, 4-Carboxyphenylboronic acid 30418-59-8, m-Aminophenylboronic acid 101084-81-5 RL: ARU (Analytical role, unclassified); ANST (Analytical study) (determination of percentage glycated Hb) RN98-80-6 HCAPLUS CNBoronic acid, phenyl- (9CI) (CA INDEX NAME)

RN 14047-29-1 HCAPLUS CN Benzoic acid, 4-borono- (9CI) (CA INDEX NAME)

RN 30418-59-8 HCAPLUS
CN Boronic acid, (3-aminophenyl)- (9CI) (CA INDEX NAME)

RN 101084-81-5 HCAPLUS CN Benzoic acid, 3-borono-5-nitro- (6CI, 9CI) (CA INDEX NAME)



REFERENCE COUNT:

THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 20 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1998:482779 HCAPLUS

DOCUMENT NUMBER:

129:227130

TITLE:

Bioluminescent properties of fluorinated

semi-synthetic aequorins

AUTHOR (S):

Hirano, Takashi; Ohmiya, Yoshihiro; Maki, Shojiro;

Niwa, Haruki; Ohashi, Mamoru

CORPORATE SOURCE:

Department of Applied Physics and Chemistry, The University of Electro-Communications, Tokyo, 182,

Japan

SOURCE:

Tetrahedron Letters (1998), 39(31), 5541-5544

CODEN: TELEAY; ISSN: 0040-4039

PUBLISHER:

Elsevier Science Ltd.

DOCUMENT TYPE:

Journal English

LANGUAGE:

Bioluminescent properties of semi-synthetic aequorins containing coelenterazine analogs I (X, Y = H, F) possessing fluoro group(s) on the 6-(4-hydroxyphenyl) group match the fluorescent behavior of the phenolate anions of the corresponding fluorinated coelenteramide analogs II. This indicates that the phenolate anion of coelenteramide is the light-emitter in aequorin bioluminescence.

6-3 (General Biochemistry) CC

Section cross-reference(s): 28

coelenterazine contq aequorin prepn bioluminescence spectra; coelenteramide contq aequorin prepn bioluminescence spectra; fluorescence chemiluminescence spectra coelenterazine deriv; chemiluminescence fluorescence spectrà coelenteramide deriv

IT Aequorins

> RL: BPR (Biological process); BSU (Biological study, unclassified); PRP (Properties); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation); PROC (Process)

(apoaequorins, coelenterazine- or coelenteramide-conjugates; preparation and bioluminescent properties of fluorinated semi-synthetic aeguorins)

IT Chromophores

(fluorinated; preparation and bioluminescent properties of fluorinated semi-synthetic aequorins)

TТ Fluorescence

> Luminescence, bioluminescence Luminescence, chemiluminescence

(preparation and bioluminescent properties of fluorinated semi-synthetic aequorins)

ITAequorins

RL: BPR (Biological process); BSU (Biological study, unclassified); PRP

(Properties); BIOL (Biological study); PROC (Process)
 (preparation and bioluminescent properties of fluorinated
 semi-synthetic aequorins)

IT 212842-91-6P 212842-92-7P

RL: BPR (Biological process); BSU (Biological study, unclassified); PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation); PROC (Process); RACT (Reactant or reagent)

(preparation and bioluminescent properties of fluorinated semi-synthetic aequorins)

IT 50611-86-4P 55779-48-1P 212842-91-6DP, apoaequorin conjugates 212842-92-7DP, apoaequorin conjugates 212842-94-9P 212842-96-1P RL: BPR (Biological process); BSU (Biological study, unclassified); PRP (Properties); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation); PROC (Process)

(preparation and **bioluminescent** properties of fluorinated semi-synthetic aequorins)

IT 62790-85-6 **133057-83-7 156635-87-9** 174680-55-8

RL: RCT (Reactant); RACT (Reactant or reagent)

(preparation and **bioluminescent** properties of fluorinated semi-synthetic aequorins)

IT 165330-28-9P 212842-97-2P 212842-99-4P 212843-00-0P 212843-01-1P RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and **bioluminescent** properties of fluorinated semi-synthetic aequorins)

IT 133057-83-7 156635-87-9

RL: RCT (Reactant); RACT (Reactant or reagent)
 (preparation and bioluminescent properties of fluorinated
 semi-synthetic aequorins)

RN 133057-83-7 HCAPLUS

CN Boronic acid, [3-fluoro-4-(phenylmethoxy)phenyl]- (9CI) (CA INDEX NAME)

HO-B
$$O-CH_2-Ph$$

RN 156635-87-9 HCAPLUS

CN Boronic acid, [2,3-difluoro-4-(phenylmethoxy)phenyl]- (9CI) (CA INDEX NAME)

26 THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT:

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 21 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1997:766280 HCAPLUS

DOCUMENT NUMBER:

128:99225

TITLE:

Structure-activity relationships of boronic acids

AUTHOR(S): Ji, X.; Kricka, L. J.

CORPORATE SOURCE:

Dep. Pathology & Lab. Medicine, Univ. Pennsylvania

Medical Center, Philadelphia, PA, 19104, USA Bioluminescence and Chemiluminescence: Molecular

SOURCE: Reporting with Photons, Proceedings of the International Symposium on Bioluminescence and

Chemiluminescence, 9th, Woods Hole, Mass., Oct. 4-8, 1996 (1997), Meeting Date 1996, 477-480. Editor(s): Hastings, J. W.; Kricka, L. J.; Stanley, P. E. Wiley:

Chichester, UK. CODEN: 65JYAO Conference

DOCUMENT TYPE:

LANGUAGE: English

The chemiluminescent horseradish peroxidase-catalyzed oxidation of AB luminol is enhanced by firefly luciferin, 4-substituted phenols, substituted naphthols, aromatic amines, phenylboronic acids, and a series of aromatic mols. Aryl boronic acids represent one of the latest classes of compds. that are effective as enhancers. In a effort to understand the relationship between enhancement activity and structure, an extensive screening study of a range of mono-, di-, and poly-substituted boronic acid derivs. was undertaken. The parent compound, phenylboronic acid, had no significant effect on blank or signal, in contrast to the various substituted phenylboronic acids that acted as enhancers. The majority of the 4-substituted phenylboronic acids were effective enhancers of light emission (except 4-methoxy and 4-phenoxy), but in contrast, none of the 2or 3-substituted derivs. tested enhanced light emission.

CC 7-3 (Enzymes)

peroxidase chemiluminescence enhancement boronate structure STactivity

Structure-activity relationship IT

(peroxidase chemiluminescence-enhancing; structure-activity relationships of peroxidase chemiluminescence enhancement by boronic acids)

IT Luminescence, chemiluminescence

> (structure-activity relationships of peroxidase chemiluminescence enhancement by boronic acids)

IT 98-80-6, Phenylboronic acid 1679-18-1,

> 4-Chlorophenylboronic acid 3900-89-8, 2-Chlorophenylboronic acid 4151-80-8, 4,4'-Bis (phenylboronic acid) 4406-77-3,

2-Phenyl-1,3,2-dioxaborinane 4426-47-5, 1-Butaneboronic acid

4612-26-4, 1,4-Phenyldiboronic acid 4688-76-0,

2-Biphenylboronic acid 5122-94-1, 4-Biphenylboronic acid

5122-99-6, 4-Iodophenylboronic acid 5467-74-3,

4-Bromophenylboronic acid 5720-05-8, 4-Methylphenylboronic acid 5720-07-0, 4-Methoxybenzeneboronic acid 13331-27-6,

3-Nitrophenylboronic acid 13922-41-3, 1-Naphthaleneboronic acid 16419-60-6, 2-Tolueneboronic acid 17865-11-1,

4-(Trimethylsilyl)benzeneboronic acid 23147-97-9,

Diphenylisobutoxyborane 30418-59-8, 3-Aminophenylboronic acid

63185-97-7 63503-60-6, 3-51067-38-0

Chlorophenylboronic acid 67492-50-6, 3,5-Dichlorophenylboronic acid 68716-47-2, 2,4-Dichlorophenylboronic acid

```
73852-18-3, 2,4,6-Trichlorophenylboronic acid 80500-28-3, 4-Carboxy-3-nitrophenylboronic acid 89694-45-1,
                                          5-Bromo-2-methoxybenzeneboronic acid
    {\tt 4-Bromophenyl-di-n-butoxyborane} \ \ \textbf{100124-06-9} \,,
                                105169-33-3 108847-76-3,
    4-Dibenzofuranboronic acid
    1-Thianthreneboronic acid 135145-90-3, 2,5-Dichlorophenylboronic
    acid 151169-66-3 151169-67-4, 4-Chloro-3-
    nitrophenylboronic acid 151169-69-6, 4-(4'-Bromodiphenyl)di-n-
    butoxyborane 151169-70-9, 4-Chlorophenyl-di-(4-chlorophenoxy)borane
    151169-71-0, N-(4-Chlorophenyl)-4-aminophenyl-boronic acid
    151169-74-3, 2,3-Dichlorophenylboronic acid
                                                  151169-76-5,
    Di-(3,5-dichlorophenoxy)-3,5-dichlorophenylborane 151196-37-1,
    3-Amino-2,4,6-trichlorophenylboronic acid 159896-15-8,
    trans-4-(3-Propenoic acid) phenylboronic acid
                                                   162125-08-8,
    3,4-Dichlorophenylboronic acid 173194-95-1, 6-Hydroxy-2-
    naphthaleneboronic acid 178320-31-5 201346-84-1,
    2-(Methylthiomethyl) phenylboronic acid 201346-86-3,
    3-Chloroacetylaminophenylboronic acid 201346-87-4 201347-10-6,
    Di-(3,4,6-trichlorophenoxy)-3,4,6-trichlorophenylborane
                                                             201347-12-8,
    Di-(1-naphthoxy)-1-naphthylborane 201347-14-0, 2-Bromomethylphenyl-di-(2-
    bromomethylphenoxy)borane 201347-17-3 201347-20-8
    RL: BAC (Biological activity or effector, except adverse); BSU (Biological
    study, unclassified); PRP (Properties); BIOL (Biological study)
        (structure-activity relationships of peroxidase
       chemiluminescence enhancement by boronic acids)
IT
    521-31-3, Luminol
                        7722-84-1, Hydrogen peroxide (H2O2), biological
              9003-99-0, Peroxidase
    RL: BPR (Biological process); BSU (Biological study, unclassified); BIOL
    (Biological study); PROC (Process)
        (structure-activity relationships of peroxidase
       chemiluminescence enhancement by boronic acids)
IT
    98-80-6, Phenylboronic acid 1679-18-1,
    4-Chlorophenylboronic acid 3900-89-8, 2-Chlorophenylboronic acid
    4151-80-8, 4,4'-Bis (phenylboronic acid) 4406-77-3,
    2-Phenyl-1,3,2-dioxaborinane 4612-26-4, 1,4-Phenyldiboronic acid
    4688-76-0, 2-Biphenylboronic acid 5122-94-1,
    4-Biphenylboronic acid 5122-99-6, 4-Iodophenylboronic acid
    5467-74-3, 4-Bromophenylboronic acid 5720-05-8,
    4-Methylphenylboronic acid 5720-07-0, 4-Methoxybenzeneboronic
    acid 13331-27-6, 3-Nitrophenylboronic acid 13922-41-3,
    1-Naphthaleneboronic acid 16419-60-6, 2-Tolueneboronic acid
    17865-11-1, 4-(Trimethylsilyl)benzeneboronic acid
    30418-59-8, 3-Aminophenylboronic acid 51067-38-0
    63503-60-6, 3-Chlorophenylboronic acid 67492-50-6,
    3,5-Dichlorophenylboronic acid 68716-47-2, 2,4-
    Dichlorophenylboronic acid 73852-18-3, 2,4,6-
    Trichlorophenylboronic acid 80500-28-3, 4-Carboxy-3-
    nitrophenylboronic acid 89694-45-1, 5-Bromo-2-
    methoxybenzeneboronic acid 96983-22-1, 4-Bromophenyl-di-n-
    butoxyborane 100124-06-9, 4-Dibenzofuranboronic acid
    108847-76-3, 1-Thianthreneboronic acid 135145-90-3,
    2,5-Dichlorophenylboronic acid 151169-66-3 151169-67-4
    , 4-Chloro-3-nitrophenylboronic acid 151169-69-6,
    4-(4'-Bromodiphenyl)di-n-butoxyborane 151169-71-0,
    N-(4-Chlorophenyl)-4-aminophenyl-boronic acid 151169-74-3,
    2,3-Dichlorophenylboronic acid 151196-37-1, 3-Amino-2,4,6-
    trichlorophenylboronic acid 159896-15-8, trans-4-(3-Propenoic
    acid) phenylboronic acid 173194-95-1, 6-Hydroxy-2-
    naphthaleneboronic acid 178320-31-5 201346-84-1,
```

2-(Methylthiomethyl)phenylboronic acid 201346-86-3, 3-Chloroacetylaminophenylboronic acid 201346-87-4

201347-17-3 201347-20-8

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); PRP (Properties); BIOL (Biological study)

(structure-activity relationships of peroxidase chemiluminescence enhancement by boronic acids)

RN98-80-6 HCAPLUS

Boronic acid, phenyl- (9CI) (CA INDEX NAME) CN

1679-18-1 HCAPLUS RN

Boronic acid, (4-chlorophenyl) - (9CI) (CA INDEX NAME) CN

3900-89-8 HCAPLUS RN

Boronic acid, (2-chlorophenyl) - (9CI) (CA INDEX NAME) CN

4151-80-8 HCAPLUS RN

Boronic acid, [1,1'-biphenyl]-4,4'-diylbis- (9CI) (CA INDEX NAME) CN

4406-77-3 HCAPLUS RN

CN 1,3,2-Dioxaborinane, 2-phenyl- (9CI) (CA INDEX NAME)

RN 4612-26-4 HCAPLUS

CN Boronic acid, 1,4-phenylenebis- (9CI) (CA INDEX NAME)

RN 4688-76-0 HCAPLUS

CN Boronic acid, [1,1'-biphenyl]-2-yl- (9CI) (CA INDEX NAME)

RN 5122-94-1 HCAPLUS

CN Boronic acid, [1,1'-biphenyl]-4-yl- (9CI) (CA INDEX NAME)

RN 5122-99-6 HCAPLUS

CN Boronic acid, (4-iodophenyl) - (9CI) (CA INDEX NAME)

RN 5467-74-3 HCAPLUS CN Boronic acid, (4-bromophenyl)- (9CI) (CA INDEX NAME)

RN 5720-05-8 HCAPLUS CN Boronic acid, (4-methylphenyl)- (9CI) (CA INDEX NAME)

RN 5720-07-0 HCAPLUS CN Boronic acid, (4-methoxyphenyl)- (9CI) (CA INDEX NAME)

RN 13331-27-6 HCAPLUS CN Boronic acid, (3-nitrophenyl)- (9CI) (CA INDEX NAME)

RN 13922-41-3 HCAPLUS

CN Boronic acid, 1-naphthalenyl- (9CI) (CA INDEX NAME)

RN 16419-60-6 HCAPLUS

CN Boronic acid, (2-methylphenyl) - (9CI) (CA INDEX NAME)

RN 17865-11-1 HCAPLUS

CN Boronic acid, [4-(trimethylsilyl)phenyl]- (9CI) (CA INDEX NAME)

RN 30418-59-8 HCAPLUS

CN Boronic acid, (3-aminophenyl) - (9CI) (CA INDEX NAME)

RN 51067-38-0 HCAPLUS

CN Boronic acid, (4-phenoxyphenyl) - (9CI) (CA INDEX NAME)

RN 63503-60-6 HCAPLUS

CN Boronic acid, (3-chlorophenyl) - (9CI) (CA INDEX NAME)

RN 67492-50-6 HCAPLUS

CN Boronic acid, (3,5-dichlorophenyl) - (9CI) (CA INDEX NAME)

RN 68716-47-2 HCAPLUS

CN Boronic acid, (2,4-dichlorophenyl) - (9CI) (CA INDEX NAME)

RN 73852-18-3 HCAPLUS

CN Boronic acid, (2,4,6-trichlorophenyl) - (9CI) (CA INDEX NAME)

RN 80500-28-3 HCAPLUS

CN Benzoic acid, 4-borono-2-nitro- (6CI, 9CI) (CA INDEX NAME)

RN 89694-45-1 HCAPLUS

CN Boronic acid, (5-bromo-2-methoxyphenyl) - (9CI) (CA INDEX NAME)

RN 96983-22-1 HCAPLUS

CN Boronic acid, (4-bromophenyl)-, dibutyl ester (9CI) (CA INDEX NAME)

RN 100124-06-9 HCAPLUS CN Boronic acid, 4-dibenzofuranyl- (9CI) (CA INDEX NAME)

RN 108847-76-3 HCAPLUS CN Boronic acid, 1-thianthrenyl- (9CI) (CA INDEX NAME)

RN 135145-90-3 HCAPLUS CN Boronic acid, (2,5-dichlorophenyl) - (9CI) (CA INDEX NAME)

RN 151169-66-3 HCAPLUS
CN Boronic acid, [2-hydroxy-5-[[3-(trifluoromethyl)phenyl]azo]phenyl]- (9CI)
(CA INDEX NAME)

RN 151169-67-4 HCAPLUS

CN Boronic acid, (4-chloro-3-nitrophenyl) - (9CI) (CA INDEX NAME)

RN 151169-69-6 HCAPLUS

CN Boronic acid, (4'-bromo[1,1'-biphenyl]-4-yl)-, dibutyl ester (9CI) (CA INDEX NAME)

RN 151169-71-0 HCAPLUS

CN Boronic acid, [4-[(4-chlorophenyl)amino]phenyl]- (9CI) (CA INDEX NAME)

RN 151169-74-3 HCAPLUS

CN Boronic acid, (2,3-dichlorophenyl) - (9CI) (CA INDEX NAME)

RN 151196-37-1 HCAPLUS

CN Boronic acid, (3-amino-2,4,6-trichlorophenyl) - (9CI) (CA INDEX NAME)

$$\begin{array}{c|c} \text{Cl} & \text{Cl} \\ \\ \text{H}_2\text{N} & \text{B-OH} \\ \\ \text{Cl} & \text{OH} \end{array}$$

RN 159896-15-8 HCAPLUS

CN 2-Propenoic acid, 3-(4-boronophenyl)-, (2E)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

RN 173194-95-1 HCAPLUS

CN Boronic acid, (6-hydroxy-2-naphthalenyl) - (9CI) (CA INDEX NAME)

RN 178320-31-5 HCAPLUS

CN Benzoic acid, 4-[(3-borono-4-hydroxyphenyl)azo]- (9CI) (CA INDEX NAME)

RN 201346-84-1 HCAPLUS

CN Boronic acid, [2-[(methylthio)methyl]phenyl]- (9CI) (CA INDEX NAME)

$$\begin{array}{c} \text{CH}_2\text{--}\,\text{SMe} \\ \\ \text{B--}\,\text{OH} \\ \\ \text{OH} \end{array}$$

RN 201346-86-3 HCAPLUS

CN Boronic acid, [3-[(chloroacetyl)amino]phenyl]- (9CI) (CA INDEX NAME)

RN 201346-87-4 HCAPLUS

CN Boronic acid, [3-[(2-methyl-1-oxobutyl)amino]phenyl]- (9CI) (CA INDEX NAME)

RN 201347-17-3 HCAPLUS

CN Boronic acid, (3-nitrophenyl)-, calcium salt (2:1) (9CI) (CA INDEX NAME)

●1/2 Ca

RN 201347-20-8 HCAPLUS

CN Boronic acid, [2-[(methylsulfinyl)methyl]phenyl]- (9CI) (CA INDEX NAME)

$$\begin{array}{c} \text{O} \\ \parallel \\ \text{CH}_2-\text{S-Me} \\ \\ \text{B-OH} \\ \mid \\ \text{OH} \end{array}$$

REFERENCE COUNT:

THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 22 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

9

ACCESSION NUMBER: DOCUMENT NUMBER:

1997:766279 HCAPLUS - 128:85737

TITLE:

Effects of polymers on the boronic acid enhanced horseradish peroxidase-luminol-hydrogen peroxide

reaction

AUTHOR(S):

Ji, X.; Kricka, L. J.

CORPORATE SOURCE:

Dep. Pathology & Lab. Medicine, Univ. Pennsylvania

Medical Center, Philadelphia, PA, 19104, USA

SOURCE:

Bioluminescence and Chemiluminescence: Molecular

Reporting with Photons, Proceedings of the International Symposium on Bioluminescence and Chemiluminescence, 9th, Woods Hole, Mass., Oct. 4-8, 1996 (1997), Meeting Date 1996, 473-476. Editor(s): Hastings, J. W.; Kricka, L. J.; Stanley, P. E. Wiley:

Chichester, UK. CODEN: 65JYAO Conference

DOCUMENT TYPE: LANGUAGE:

English

AB Soluble hydroxy-polymers stabilize light emission from boronic acid enhanced horseradish peroxidase catalyzed chemiluminescent oxidation of luminol. The results of screening studies of different hydroxy-polymers and the effects of polymer concentration and mol. weight are reported. Hydroxypropyl cellulose, poly(ethylene glycol), poly(tetramethylene ether glycol) were superior to dextran and poly(propylene glycol) in terms of light emission signal, but these polymers also increased the background light emission >2-fold. Increasing the mol. weight of the polymer had no major effect on the stabilization of the light emission for any

combination of polymer or enhancer tested. The polymer-mediated stabilization effect was concentration dependent.

CC 7-3 (Enzymes)

ST peroxidase chemiluminescence boronate polymer

IT Luminescence, chemiluminescence

(effects of polymers on the boronic acid enhanced horseradish peroxidase-luminol-hydrogen peroxide reaction)

IT 5122-94-1, 4-Biphenylboronic acid 5122-99-6,

4-Iodophenylboronic acid 9004-54-0, Dextran, biological studies 9004-64-2, Hydroxypropyl cellulose 25190-06-1, Poly(tetramethylene ether glycol) 25322-68-3, Poly(ethylene glycol) 25322-69-4, Poly(propylene glycol)

RL: ARG (Analytical reagent use); BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study); USES (Uses)

(effects of polymers on the boronic acid enhanced horseradish peroxidase-luminol-hydrogen peroxide reaction)

IT 5122-94-1, 4-Biphenylboronic acid 5122-99-6,

4-Iodophenylboronic acid

RL: ARG (Analytical reagent use); BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study); USES (Uses)

(effects of polymers on the boronic acid enhanced horseradish peroxidase-luminol-hydrogen peroxide reaction)

RN 5122-94-1 HCAPLUS

CN Boronic acid, [1,1'-biphenyl]-4-yl- (9CI) (CA INDEX NAME)

RN 5122-99-6 HCAPLUS

CN Boronic acid, (4-iodophenyl) - (9CI) (CA INDEX NAME)

REFERENCE COUNT:

5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 23 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: 1997:

1997:636220 HCAPLUS

DOCUMENT NUMBER:

127:305048

TITLE: INVENTOR(S): Acridan compounds

Akhavan-Tafti, Hashem; Arghavani, Zahra; Desilva,

Renuka

PATENT ASSIGNEE(S):

Lumigen, Inc., USA

SOURCE:

U.S., 18 pp., Cont.-in-part of U.S. Ser. No. 300,462.

CODEN: USXXAM

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT: 12

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 5670644	Α	19970923	US 1996-647383	19960509
US 5491072	Α	19960213	US 1993-61810	19930517
US 5593845	A	19970114	US 1994-205093	19940302
US 5523212	A	19960604	US 1994-228290	19940415
JP 08500125	T2	19960109	JP 1994-525766	19940516
JP 3231777	В2	20011126		
AU 9944594	A1	19991111	AU 1999-44594	19990819
AU 733635	B2	20010517		
PRIORITY APPLN. INFO.:			U\$ 1993-61810	A2 19930517
			US 1994-205093	A2 19940302
			US 1994-228290	A2 19940415
			US 1994-300462	A2 19940902
			WO 1994-US5437	W 19940516
			AU 1995-34619	A3 19950830

OTHER SOURCE(S):

MARPAT 127:305048

A chemiluminescent assay method, compns., kits, and chemiluminescent acridan compds. are described which use a 2-step chemiluminescent reaction process. The reaction involves an acridan compound, preferably a derivative of an N-alkyl acridan-9-carboxylic acid, which undergoes a reaction with a peroxide compound, a peroxidase enzyme, and an enhancer under conditions of time, temperature, and pH which permit the accumulation of an intermediate compound, which is subsequently induced to produce a burst of light by raising the pH. The result is generation of very-high-intensity light from the reaction. The peroxidase enzyme is present alone or linked to a member of a specific binding pair in an immunoassay, DNA probe assay, or other assay where the hydrolytic enzyme is bound to a reporter mol. The method is particularly amenable to automated assays because of the separation of the incubation and light-generating steps.

ICM C07D285-38 IC

> C07D295-00; G01N033-533; G01N033-532 ICS

NCL 546103000

9-14 (Biochemical Methods)

Section cross-reference(s): 3, 15, 27, 80

acridan compd prepn chemiluminescence enzymic assay; peroxidase ST detn alkyl acridancarboxylate chemiluminescence

ΤT Chemiluminescence spectroscopy

Nucleic acid hybridization

Test kits

(acridan compds. preparation for chemiluminescence assays)

IT Antibodies

Antigens

DNA

Haptens

Nucleic acids

Proteins, general, analysis

RL: ANT (Analyte); ANST (Analytical study)

```
(acridan compds. preparation for chemiluminescence assays)
TT
     Peroxides, reactions
     RL: ARG (Analytical reagent use); RCT (Reactant); ANST (Analytical study);
     RACT (Reactant or reagent); USES (Uses)
        (acridan compds. preparation for chemiluminescence assays)
     Onium compounds
IT
     RL: ARG (Analytical reagent use); SPN (Synthetic preparation); ANST
     (Analytical study); PREP (Preparation); USES (Uses)
        (acridinium; acridan compds. preparation for chemiluminescence
       assays)
IT
     Surfactants
        (anionic; acridan compds. preparation for chemiluminescence
       assays)
TT
     Immunoassay
        (chemiluminescence; acridan compds. preparation for
       chemiluminescence assays)
TT
        (enzyme; acridan compds. preparation for chemiluminescence assays)
     Surfactants
IT
        (nonionic; acridan compds. preparation for chemiluminescence
                         9035-82-9, Dehydrogenase
     9035-73-8, Oxidase
TT
     RL: ANT (Analyte); ANST (Analytical study)
        (acridan compds. preparation for chemiluminescence assays)
IT
     9003-99-0, Peroxidase
     RL: ANT (Analyte); ARG (Analytical reagent use); ANST (Analytical study);
     USES (Uses)
        (acridan compds. preparation for chemiluminescence assays)
IT
     106-41-2, p-Bromophenol
                              124-43-6
                                        135-19-3, 2-Naphthol, uses
     540-38-5, p-Iodophenol
                             719-54-0, N-Methylacridone 5122-99-6,
     4-Iodophenylboronic acid 7400-08-0, p-Hydroxycinnamic acid
                                                                   7632-04-4,
     Sodium perborate
                       7722-84-1, Hydrogen peroxide, uses
                                                            15231-91-1,
     6-Bromo-2-naphthol
                         130897-36-8
                                       172834-33-2
                                                     172834-43-4
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (acridan compds. preparation for chemiluminescence assays)
     92-81-9DP, Acridan, derivs. 177535-21-6P 177535-23-8P
                                                                177535-24-9P
     177535-25-0P
                    197156-16-4P
                                  197156-17-5P
                                                 197156-18-6P
                                                                197156-19-7P
     197156-20-0P
                   197156-35-7P
                                  197156-36-8DP, N-alkyl
                                                          197256-32-9P
     RL: ARG (Analytical reagent use); SPN (Synthetic preparation); ANST
     (Analytical study); PREP (Preparation); USES (Uses)
        (acridan compds. preparation for chemiluminescence assays)
IT
     60-00-4, EDTA, analysis
     RL: ARU (Analytical role, unclassified); ANST (Analytical study)
        (acridan compds. preparation for chemiluminescence assays)
     79-37-8, Oxalyl chloride 95-78-3, 2,5-Dimethylaniline 98-59-9,
IT
     p-Toluenesulfonyl chloride
                                101-16-6, 3-Methoxydiphenylamine 101-17-7,
                           102-56-7, 2,5-Dimethoxyaniline
     3-Chlorodiphenylamine
                                                            108-95-2, Phenol,
                333-27-7, Methyl triflate 371-42-6, 4-Fluorothiophenol
     reactions
     1205-64-7, 3-Methyldiphenylamine 2398-37-0, 3-Bromoanisole
                                                                   3467-59-2
     33264-65-2
                 50868-72-9, 5-Methoxy-2-methylaniline
                                                         92248-06-1,
     Bis (3-methoxyphenyl) amine
                               113798-74-6, 2,3,6-Trifluorophenol
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (acridan compds. preparation for chemiluminescence assays)
     2050-44-4P, 2,5-Dimethylacetanilide 32446-14-3P 42595-25-5P,
IT
     3-Chloroacridine-9-carboxylic acid 50868-75-2P
                                                       130266-60-3P,
     3-Methylacridine-9-carboxylic acid
                                         154471-37-1P, 1-Methylacridine-9-
     carboxylic acid 172834-71-8P
                                    177535-29-4P
                                                   177535-32-9P
                                 177535-38-5P
                                                 177535-40-9P
     177535-33-0P
                  177535-37-4P
                                                                177535-41-0P
     177535-42-1P 177535-44-3P 177535-45-4P
                                                 178920-79-1P
                                                                197156-21-1P
```

```
197156-24-4P
                                                  197156-25-5P
                                                                 197156-26-6P
     197156-22-2P
                    197156-23-3P
                                                                 197156-31-3P
                                   197156-29-9P
                                                  197156-30-2P
    197156-27-7P
                   197156-28-8P
                                                  197256-33-0P
                                                                 197256-34-1P
                   197156-33-5P
                                   197156-34-6P
     197156-32-4P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (acridan compds. preparation for chemiluminescence assays)
                                  177535-46-5P
                    177535-43-2P
IT
     172834-72-9P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (acridan compds. preparation for chemiluminescence assays)
     5122-99-6, 4-Iodophenylboronic acid
IT
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (acridan compds. preparation for chemiluminescence assays)
     5122-99-6 HCAPLUS
RN
     Boronic acid, (4-iodophenyl) - (9CI) (CA INDEX NAME)
CN
```

L28 ANSWER 24 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1997:560112 HCAPLUS

DOCUMENT NUMBER:

127:244590

TITLE:

Effect of polymers on enhanced

chemiluminescent assays for peroxidase and

peroxidase labels

AUTHOR(S):

Ji, Xiaoying; Kricka, Larry J.

CORPORATE SOURCE:

Department of Pathology and Laboratory Medicine,

University of Pennsylvania Medical Center,

Philadelphia, PA, 19104, USA

SOURCE:

Journal of Bioluminescence and Chemiluminescence

(1996), 11(6), 303-307

CODEN: JBCHE7; ISSN: 0884-3996

PUBLISHER: DOCUMENT TYPE:

Wiley Journal English LANGUAGE:

Hydroxypropyl methylcellulose, hydroxyethyl cellulose, and hydroxybutyl methylcellulose stabilized light emission in a boronic acid-enhanced chemiluminescent assay for horseradish peroxidase. stabilization of light emission was concentration-dependent and more effective with substituted boronic acid enhancers (e.g. 4-iodophenylboronic acid) than with substituted phenol enhancers (e.g. 4-iodophenol). Hydroxybutyl methylcellulose improved the linearity of the dose-response curve in a peroxidase-based antioxidant assay and stabilized light emission post-consumption of the antioxidant (Trolox). This polymer had no effect on the signal from a peroxidase label immobilized on a membrane (dot blot) or on the inside surface of a microwell in an enzyme immunoassay for TSH. 7-1 (Enzymes)

CC

cellulose effect chemiluminescent assay peroxidase ST

IT Immunoassay

(enzyme; effect of polymers on enhanced chemiluminescent assays for peroxidase and peroxidase labels)

9002-71-5, Thyrotropin 9003-99-0, Peroxidase TΤ

RL: ANT (Analyte); ANST (Analytical study) (effect of polymers on enhanced chemiluminescent assays for peroxidase and peroxidase labels) peroxidase and peroxidase labels) IT 540-38-5, 4-Iodophenol 5122-94-1, 4-Biphenylboronic acid 5122-99-6, 4-Iodophenylboronic acid 5467-74-3, 4-Bromophenylboronic acid 7400-08-0, 4-Hydroxycinnamic acid 9004-62-0, Hydroxyethyl cellulose 9004-65-3, Hydroxypropyl methylcellulose 9041-56-9, Hydroxybutyl methylcellulose 53188-07-1, Trolox 159896-15-8 RL: ARU (Analytical role, unclassified); ANST (Analytical study) (effect of polymers on enhanced chemiluminescent assays for peroxidase and peroxidase labels) 5122-94-1, 4-Biphenylboronic acid 5122-99-6, IT 4-Iodophenylboronic acid 5467-74-3, 4-Bromophenylboronic acid RL: ARU (Analytical role, unclassified); ANST (Analytical study) (effect of polymers on enhanced chemiluminescent assays for peroxidase and peroxidase labels) RN5122-94-1 HCAPLUS Boronic acid, [1,1'-biphenyl]-4-yl- (9CI) (CA INDEX NAME) CN

RN 5122-99-6 HCAPLUS CN Boronic acid, (4-iodophenyl)- (9CI) (CA INDEX NAME)

RN 5467-74-3 HCAPLUS CN Boronic acid, (4-bromophenyl)- (9CI) (CA INDEX NAME)

RN159896-15-8 HCAPLUS 2-Propenoic acid, 3-(4-boronophenyl)-, (2E)- (9CI) (CA INDEX NAME) CN

Double bond geometry as shown.

REFERENCE COUNT:

THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS 12 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 25 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1997:360371 HCAPLUS

DOCUMENT NUMBER:

127:92085

TITLE:

Synergistic enhancement of the horseradish peroxidase-catalyzed oxidation of luminol by

4-substituted phenylobronic acids Kricka, Larry J.; Ji, Xiaoying

AUTHOR(S):

CORPORATE SOURCE:

Dep. Pathology and Lab. Medicine, Univ. Pennsylvania

Medical Center, Philadelphia, PA, 19104, USA

SOURCE:

Talanta (1997), 44(6), 1073-1079 CODEN: TLNTA2; ISSN: 0039-9140

PUBLISHER:

Elsevier Journal

DOCUMENT TYPE:

English

LANGUAGE: Combinations of 4-substituted phenylboronic acids [phenyl, iodo, bromo, and trans-4-(3-propenoic acid) substituents] have been discovered to have synergistic effects in the horseradish peroxidase (HRP) catalyzed chemiluminescent oxidation of luminol. Three types of effect have been observed: (1) synergistic reduction in the background light emission of a luminol-peroxide assay reagent to a value lower than the background obtained with either enhancer individually; (2) increase in signal to background ratio (S/B) in the presence of HRP to a value higher than the S/B obtained with either enhancer individually (synergy) or to a value higher than the combined S/B obtained with each enhancer (synergistic enhancement); and (3) for some combinations of enhancers, an increase in signal in the presence of HRP to a value higher than the signal obtained with either enhancer individually (synergy), or to a value higher than the combined signal obtained with each enhancer (synergistic enhancement). The magnitude of the effect was moderate but the synergistic decreases in background and increased in signal produced increases in S/B up to four-fold. Examples of synergistic pairs of enhancers included 4-biphenyl and 4-bromophenylboronic acid; 4-biphenyl and 4-iodophenylboronic acid; and trans-4-(3-propenoic acid) and 4-iodophenylboronic acid. Generally, synergy was obtained at several concns. of all of the combinations of enhancers tested, and at different time points in the reaction due to the different light emission kinetics of the enhanced reactions. The mechanism of this synergistic effect has not been elucidated but may involve the enhancers acting at different points in the complex chemiluminescent peroxidase catalyzed oxidation reaction.

CC7-4 (Enzymes)

TT

5122-94-1, 4-Biphenylboronic acid 5122-99-6,

4-Iodophenylboronic acid **5467-74-3**, 4-Bromophenylboronic acid 9003-99-0, Peroxidase **192137-70-5**

RL: BAC (Biological activity or effector, except adverse); BPR (Biological process); BSU (Biological study, unclassified); BIOL (Biological study); PROC (Process)

(synergistic enhancement of the horseradish peroxidase-catalyzed oxidation of luminol by 4-substituted phenylobronic acids)

5122-94-1, 4-Biphenylboronic acid 5122-99-6,

4-Iodophenylboronic acid **5467-74-3**, 4-Bromophenylboronic acid **192137-70-5**

RL: BAC (Biological activity or effector, except adverse); BPR (Biological process); BSU (Biological study, unclassified); BIOL (Biological study); PROC (Process)

(synergistic enhancement of the horseradish peroxidase-catalyzed oxidation of luminol by 4-substituted phenylobronic acids)

RN 5122-94-1 HCAPLUS

CN Boronic acid, [1,1'-biphenyl]-4-yl- (9CI) (CA INDEX NAME)

IT

RN 5122-99-6 HCAPLUS

CN Boronic acid, (4-iodophenyl) - (9CI) (CA INDEX NAME)

RN 5467-74-3 HCAPLUS

CN Boronic acid, (4-bromophenyl) - (9CI) (CA INDEX NAME)

RN 192137-70-5 HCAPLUS

CN 2-Propenoic acid, 3-(4'-borono[1,1'-biphenyl]-4-yl)-, (E)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

REFERENCE COUNT:

THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS 12 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 26 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1996:570608 HCAPLUS

DOCUMENT NUMBER:

125:268952

TITLE:

Synthesis and characterization of 4-iodophenylboronic

acid: a new enhancer for the horseradish peroxidase-catalyzed chemiluminescent

oxidation of luminol

AUTHOR(S):

Kricka, Larry J.; Cooper, Mark; Ji, Xiaoying

CORPORATE SOURCE:

Dep. Pathology, Univ. Pennsylvania Med. Center,

Philadelphia, PA, 19104, USA

SOURCE:

Analytical Biochemistry (1996), 240(1), 119-125

CODEN: ANBCA2; ISSN: 0003-2697

PUBLISHER:

Academic

DOCUMENT TYPE:

Journal

LANGUAGE:

English

4-Iodophenylboronic acid has been synthesized and shown to be a potent enhancer of the chemiluminescent horseradish peroxidase (Type VI-A)-catalyzed oxidation of luminol and isoluminol. The enhancer was effective (>100-fold enhancement) in the concentration range 10-1500 μ M. Light emission in the presence of the enhancer peaked at 5-10 min after initiation of the reaction and then decayed very slowly. 4-Iodophenylboronic acid also enhanced reactions catalyzed by horseradish peroxidase Type IX. The detection limit for type VI-A horseradish peroxidase was 509 amol, and optimum signal enhancement was obtained at 769 μM compared to 154 μM for 4-iodophenol under the same conditions.

7-3 (Enzymes) CC

Section cross-reference(s): 9

iodophenylboronate prepn enhancer peroxidase luminol ST chemiluminescence; isoluminol peroxidase chemiluminescence enhancer iodophenylboronate prepn

IT Luminescence, chemi-

(synthesis and characterization of 4-iodophenylboronic acid as an enhancer for the horseradish peroxidase-catalyzed chemiluminescent oxidation of luminol and isoluminol)

ITTransferrins

RL: ANT (Analyte); ANST (Analytical study)

(synthesis and characterization of 4-iodophenylboronic acid as an enhancer for the horseradish peroxidase-catalyzed chemiluminescent oxidation of luminol and isoluminol)

IT 540-38-5, 4-Iodophenol 5467-74-3, 4-Bromophenylboronic acid RL: ARG (Analytical reagent use); BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study); USES (Uses)

(comparison; synthesis and characterization of 4-iodophenylboronic acid as an enhancer for the horseradish peroxidase-catalyzed

chemiluminescent oxidation of luminol and isoluminol)

IT 5122-99-6P, 4-Iodophenylboronic acid

RL: ARG (Analytical reagent use); BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); SPN (Synthetic preparation); ANST (Analytical study); BIOL (Biological study); PREP (Preparation); USES (Uses)

(synthesis and characterization of 4-iodophenylboronic acid as an enhancer for the horseradish peroxidase-catalyzed

chemiluminescent oxidation of luminol and isoluminol)

IT 521-31-3, Luminol 3682-14-2, Isoluminol 7722-84-1, Hydrogen peroxide, biological studies 9003-99-0, Peroxidase

RL: ARG (Analytical reagent use); BPR (Biological process); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study); PROC (Process); USES (Uses)

(synthesis and characterization of 4-iodophenylboronic acid as an enhancer for the horseradish peroxidase-catalyzed

chemiluminescent oxidation of luminol and isoluminol)

IT 121-43-7, Trimethylborate 624-38-4, 1,4-Diiodobenzene

RL: RCT (Reactant); RACT (Reactant or reagent)

(synthesis and characterization of 4-iodophenylboronic acid as an enhancer for the horseradish peroxidase-catalyzed

chemiluminescent oxidation of luminol and isoluminol)

IT 114448-26-9P, 4-Iodophenyl lithium

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(synthesis and characterization of 4-iodophenylboronic acid as an enhancer for the horseradish peroxidase-catalyzed

chemiluminescent oxidation of luminol and isoluminol)

IT 5467-74-3, 4-Bromophenylboronic acid

RL: ARG (Analytical reagent use); BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); ANST (Analytical study); BIOL (Biological study); USES (Uses)

(comparison; synthesis and characterization of 4-iodophenylboronic acid as an enhancer for the horseradish peroxidase-catalyzed chemiluminescent oxidation of luminol and isoluminol)

RN 5467-74-3 HCAPLUS

CN Boronic acid, (4-bromophenyl) - (9CI) (CA INDEX NAME)

TT 5122-99-6P, 4-Iodophenylboronic acid
RL: ARG (Analytical reagent use); BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); SPN (Synthetic preparation); ANST (Analytical study); BIOL (Biological study); PREP

(Preparation); USES (Uses)

(synthesis and characterization of 4-iodophenylboronic acid as an enhancer for the horseradish peroxidase-catalyzed

chemiluminescent oxidation of luminol and isoluminol)

RN5122-99-6 HCAPLUS

CNBoronic acid, (4-iodophenyl) - (9CI) (CA INDEX NAME)

L28 ANSWER 27 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1996:540546 HCAPLUS

DOCUMENT NUMBER:

125:212800

TITLE:

Super-sensitive enzyme immunoassay for thyroid

stimulating hormone using a new synergistic enhanced

chemiluminescent endpoint

AUTHOR (S):

Kricka, Larry J.; Ji, Xiaoying

CORPORATE SOURCE:

Medical Center, University of Pennsylvania,

Philadelphia, PA, 19104, USA

SOURCE:

Journal of Bioluminescence and Chemiluminescence

(1996), 11(3), 137-147

CODEN: JBCHE7; ISSN: 0884-3996

PUBLISHER:

Wiley Journal

DOCUMENT TYPE: LANGUAGE:

English

The enhancers 1,1'-biphenyl-4-yl boronic acid and 4-iodophenol act synergistically in the horseradish peroxidase-catalyzed oxidation of luminol. This concentration-dependent effect reduces background, increases signal and hence improves signal/background for detection of peroxidase. The same type of synergistic effect was found when 1,1'-biphenyl-4-yl boronic acid was added to a com. enhanced chemiluminescence signal reagent (Amerlite Signal Reagent). This synergistic enhanced chemiluminescent endpoint (Amerlite Signal Reagent containing 1,1'-biphenyl-4-yl boronic acid) for a horseradish peroxidase label has been tested in the Amerlite TSH and the Amerlite TSH-30 Ultrasensitive assays. The detection limit (mean of 20 replicates of the zero standard + 2SD) in the Amerlite TSH assay was 0.0029 mlU/L, and in the Amerlite TSH-30 Ultrasensitive assay the detection limit was 0.0005 mlU/L using the synergistic enhanced endpoint. Reassessment of the detection limit using a 1:40 dilution of the first standard (0.119 mlU/L) as the lowest assay

a value of 0.0015 mlU/L for the Amerlite TSH-30 Ultrasensitive assay with the synergistic endpoint. A limited method comparison using samples from euthyroid, hyperthyroid and hypothyroid patients revealed excellent correlation between the conventional and synergistic TSH immunoassays.

2-1 (Mammalian Hormones)

ST TSH detn EIA synergistic chemiluminescent endpoint

ITBlood analysis

Hyperthyroidism

Hypothyroidism

(TSH determination in human blood by Amerlite TSH and Amerlite TSH-30

Ultrasensitive EIA with synergistic enhanced chemiluminescent endpoint)

IT 9002-71-5, Thyroid-stimulating hormone

RL: ANT (Analyte); ANST (Analytical study)

(TSH determination in human blood by Amerlite TSH and Amerlite TSH-30 Ultrasensitive EIA with synergistic enhanced chemiluminescent endpoint)

IT 5122-94-1

.: 4

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (TSH determination in human blood by Amerlite TSH and Amerlite TSH-30 Ultrasensitive EIA with synergistic enhanced chemiluminescent endpoint)

IT 5122-94-1

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (TSH determination in human blood by Amerlite TSH and Amerlite TSH-30 Ultrasensitive EIA with synergistic enhanced chemiluminescent endpoint)

RN 5122-94-1 HCAPLUS

CN Boronic acid, [1,1'-biphenyl]-4-yl- (9CI) (CA INDEX NAME)

L28 ANSWER 28 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1996:350342 HCAPLUS 125:29590

DOCUMENT NUMBER: TITLE:

Chemiluminescent assay utilizing an acridan

and peroxidase

INVENTOR(S):

Akhaven-Tafti, Hashem; Arghavani, Zahra; Desilva,

Renuka

PATENT ASSIGNEE(S):

Lumigen, Inc., USA

SOURCE: PCT Int. Appl., 52 pp.

CODEN: PIXXD2

DOCUMENT TYPE:

Patent English

LANGUAGE:
FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

ATENT NO.	KIND DATE	APPLICATION NO.	DATE			
9607912	A1 19960314	WO 1995-US11031	19950830			
W: AU, CA, CN,	FI, JP, KR					
RW: AT, BE, CH,	DE, DK, ES, FR,	GB, GR, IE, IT, LU, MC,	NL, PT, SE			
A 2197669	AA 19960314	CA 1995-2197669	19950830			
J 9534619	A1 19960327	AU 1995-34619	19950830			
778946	A1 19970618	EP 1995-931030	19950830			
778946	B1 20021023					
R: AT, BE, CH,	DE, DK, ES, FR,	GB, GR, IE, IT, LI, NL,	SE			
J 1161083	A 19971001	CN 1995-195266	19950830			
2 10508191	T2 19980818	JP 1995-509567	19950830			
	W: AU, CA, CN, RW: AT, BE, CH, A 2197669 J 9534619 P 778946	D 9607912 A1 19960314 W: AU, CA, CN, FI, JP, KR RW: AT, BE, CH, DE, DK, ES, FR, A 2197669 AA 19960314 J 9534619 A1 19960327 P 778946 A1 19970618 P 778946 B1 20021023 R: AT, BE, CH, DE, DK, ES, FR, N 1161083 A 19971001	D 9607912 A1 19960314 WO 1995-US11031 W: AU, CA, CN, FI, JP, KR RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, A 2197669 AA 19960314 CA 1995-2197669 U 9534619 A1 19960327 AU 1995-34619 P 778946 A1 19970618 EP 1995-931030 P 778946 B1 20021023 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LI, NL, U 1161083 A 19971001 CN 1995-195266			

```
AT 226728
                          Ε
                                20021115
                                            AT 1995-931030
                                                                   19950830
    AU 9944594
                         Α1
                                19991111
                                           AU 1999-44594
                                                                   19990819
    AU 733635
                         B2
                                20010517
                                            US 1994-300462
                                                                A 19940902
PRIORITY APPLN. INFO.:
                                            AU 1995-34619
                                                                A3 19950830
                                            WO 1995-US11031
                                                                W 19950830
OTHER SOURCE(S):
                         MARPAT 125:29590
    A chemiluminescent assay method utilizes a 2-step
    chemiluminescent reaction involving an acridan prepared by using
    standard reactions. In particular, a N-alkylacridan-9-carboxylic acid
derivative
    undergoes a reaction with a peroxide compound, a peroxidase enzyme and an
    enhancer, which permit the accumulation of an intermediate which is
    subsequently induced to produce a burst of light by raising the pH.
                                                                          The
    result is a generation of very high intensity light from the reaction.
    The peroxidase enzyme is present alone or linked to a member of a specific
    binding pair in an immunoassay, DNA probe assay or other assay where the
    hydrolytic enzyme is bound to a reporter mol. The method is particularly
    amenable to automated assay because of the separation of the incubation and
    light generating steps.
    ICM G01N033-535
IC
    ICS C07D219-04
    9-5 (Biochemical Methods)
CC
    Section cross-reference(s): 27
    acridan peroxidase peroxide chemiluminescence prepn
ST
IT
    Spectrochemical analysis
        (chemiluminescence, chemiluminescent assay
        utilizing acridan compound and peroxidase)
    106-41-2, p-Bromophenol
                              124-43-6, Urea peroxide
                                                        135-19-3, 2-Naphthol,
IT
    uses
            540-38-5, p-Iodophenol 5122-99-6, 4-Iodophenylboronic
            7400-08-0, p-Hydroxycinnamic acid
                                                7722-84-1, Hydrogen peroxide,
    acid
                                   15231-91-1, 6-Bromo-2-naphthol
           9003-99-0, Peroxidase
    uses
    RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (chemiluminescent assay utilizing acridan compound and
       peroxidase)
ΙT
    172834-40-1P
                   177535-19-2P
                                   177535-20-5P
                                                  177535-21-6P
                                                                 177535-22-7P
    177535-23-8P
                   177535-24-9P
                                   177535-25-0P
    RL: ARG (Analytical reagent use); RCT (Reactant); SPN (Synthetic
    preparation); ANST (Analytical study); PREP (Preparation); RACT (Reactant
    or reagent); USES (Uses)
        (chemiluminescent assay utilizing acridan compound and
       peroxidase)
                                       371-42-6, 4-Fluorothiophenol
IT
    101-16-6, 3-Methoxydiphenylamine
                 113798-74-6, 2,3,6-Trifluorophenol
                                                     130266-60-3
    42595-25-5
                                177535-30-7
    154471-37-1
                 173407-17-5
                                              177535-31-8
                                                           177535-42-1
    RL: RCT (Reactant); RACT (Reactant or reagent)
        (chemiluminescent assay utilizing acridan compound and
       peroxidase)
IT
                   172834-70-7P
                                  172834-71-8P
                                                  172834-72-9P
                                                                 177535-26-1P
    172834-54-7P
                                  177535-29-4P
    177535-27-2P
                   177535-28-3P
                                                  177535-32-9P
                                                                 177535-34-1P
                   177535-36-3P
                                  177535-37-4P
                                                  177535-39-6P
                                                                 177535-40-9P
    177535-35-2P
                   177535-43-2P
                                  177535-44-3P
                                                 177535-45-4P
                                                                 177535-46-5P
    177535-41-0P
    RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (chemiluminescent assay utilizing acridan compound and
       peroxidase)
    5122-99-6, 4-Iodophenylboronic acid
IT
    RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
```

(chemiluminescent assay utilizing acridan compound and

peroxidase)

5122-99-6 HCAPLUS RN

Boronic acid, (4-iodophenyl) - (9CI) CN(CA INDEX NAME)

L28 ANSWER 29 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1996:323929 HCAPLUS

DOCUMENT NUMBER:

125:53012

TITLE:

Enhancement of chemiluminescent reactions

INVENTOR(S):

Kricka, Larry J.

PATENT ASSIGNEE(S):

British Technology Group Limited, UK

SOURCE:

U.S., 18 pp., Cont.-in-part of U.S. Ser. No. 41,233,

abandoned.

CODEN: USXXAM

DOCUMENT TYPE:

Patent English

LANGUAGE:

2

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 5512451	Α	19960430	US 1994-217845	19940325
AU 9463827	A1	19941024	AU 1994-63827	19940331
EP 692033	A1	19960117	EP 1994-911267	19940331
EP 692033	B1	19980311		
R: BE, CH, DE,	ES, FR	, IE, IT, I	LI, NL, SE	
JP 08508342	T2	19960903	JP 1994-521849	19940331
US 5629168	Α	19970513	US 1995-478336	19950607
FI 9504653	A	19950929	FI 1995-4653	19950929
PRIORITY APPLN. INFO.:			GB 1993-6888	19930401
			US 1993-41233	19930401
			US 1992-833153	19920210
			US 1992-925689	19920807
			US 1993-157504	19931126
			WO 1994-GB700	19940331

OTHER SOURCE(S): MARPAT 125:53012

In enhanced chemiluminescent (ECL) reactions of a fused aromatic diacyl cyclic hydrazide such as luminol, a peroxidase enzyme catalyst, an oxidant such as hydrogen peroxide, and an enhancer, it is advantageous to use a combination of an organoboron enhancer such as 4-biphenylboronic acid with a non-boron-containing enhancer, especially a phenolic or aromatic amine

enhancer, particularly 4-iodophenol. ECL reactions are useful in diagnostic assays.

ICM C12Q001-28 ICS C07H015-00 IC

NCL435028000

9-5 (Biochemical Methods) CC

Section cross-reference(s): 1, 3, 73, 80

```
chemiluminescence reaction organoboron phenolic enhancer
ST
     peroxidase; immunoassay chemiluminescence reaction enhancer
     Blood analysis
TΤ
     Immunoassay
     Nucleic acid hybridization
        (chemiluminescent reactions enhancement with organoboron and
        phenolic or amine compds.)
IT
     Antibodies
     Antigens
     RL: ANT (Analyte); ANST (Analytical study)
        (chemiluminescent reactions enhancement with organoboron and
        phenolic or amine compds.)
IT
     Amines, uses
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (chemiluminescent reactions enhancement with organoboron and
        phenolic or amine compds.)
IT
     Phenols, uses
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (chemiluminescent reactions enhancement with organoboron and
        phenolic or amine compds.)
     Spectrochemical analysis
TT
        (chemiluminescence, chemiluminescent reactions
        enhancement with organoboron and phenolic or amine compds.)
IT
     Immunoassay
        (enzyme, chemiluminescent reactions enhancement with
        organoboron and phenolic or amine compds.)
TT
     Immunoassay
        (enzyme-linked immunosorbent assay, chemiluminescent
        reactions enhancement with organoboron and phenolic or amine compds.)
     7722-84-1, Hydrogen peroxide, analysis 9002-71-5, TSH
IT:
     RL: ANT (Analyte); ANST (Analytical study)
        (chemiluminescent reactions enhancement with organoboron and
        phenolic or amine compds.)
     9003-99-0, Peroxidase
TT
     RL: ANT (Analyte); ARG (Analytical reagent use); ANST (Analytical study);
     USES (Uses)
        (chemiluminescent reactions enhancement with organoboron and
        phenolic or amine compds.)
     92-04-6, 2-Chloro-4-phenylphenol
                                        92-69-3, 4-Phenylphenol
TT
     3,4-Dichlorophenol 98-54-4, 4-tert-Butylphenol 101-53-1,
     4-Benzylphenol
                    103-90-2, 4-Acetamidophenol 104-94-9 106-41-2,
                     106-44-5, 4-Methylphenol, uses 106-48-9, 4-Chlorophenol
     4-Bromophenol
                                  135-19-3, Naphth-2-ol, uses
     120-83-2, 2,4-Dichlorophenol
     N, N, N', N'-Tetramethylbenzidine 521-31-3, Luminol 540-38-5,
     4-Iodophenol 573-97-7, 1-Bromonaphth-2-ol 831-82-3, 4-Phenoxyphenol
     1634-82-8 1679-18-1, 4-Chlorophenylboronic acid
                                                       1689-82-3,
                                    2664-63-3 2975-55-5 3558-83-6,
3964-56-5, 4-Bromo-2-chlorophenol
     4-Phenylazophenol
                        1965-09-9
                                   2664-63-3
     4-4'-Hydroxyphenylbenzophenone
     4426-21-5, Diphenylborinic anhydride 5122-94-1,
     4-Biphenylboronic acid 5122-99-6, 4-Iodophenylboronic acid
     5467-74-3, p-Bromophenylboronic acid 7400-08-0,
                            7440-42-8D, Boron, organic compds.
                                                                   10041-02-8,
     4-Hydroxycinnamic acid
                            15015-57-3, 4-Hydroxyphenyl disulfide
     4-Imidazol-1-ylphenol
     15231-91-1, 6-Bromo-2-naphthol 16239-18-2, 1,6-Dibromonaphth-2-ol
     17865-11-1, 4-Trimethylsilylbenzeneboronic acid 23147-97-9
                                                                  39349-73-0,
                23795-02-0, Ethyl 3-4-hydroxyphenylpropionate
     23567-67-1
                             83474-09-3 92681-33-9 93361-16-1
     Perborate
                 46913-53-5
                115252-18-1
                                134459-06-6 143323-55-1 144774-20-9
     96983-22-1
     144774-21-0 151169-66-3 151169-67-4,
```

4-Chloro-3-nitrophenylboronic acid 151169-69-6 151169-70-9 151169-71-0 159896-15-8 178320-31-5 RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (chemiluminescent reactions enhancement with organoboron and phenolic or amine compds.) 1679-18-1, 4-Chlorophenylboronic acid 5122-94-1, IT 4-Biphenylboronic acid 5122-99-6, 4-Iodophenylboronic acid 5467-74-3, p-Bromophenylboronic acid 17865-11-1, 4-Trimethylsilylbenzeneboronic acid 96983-22-1 151169-66-3 151169-67-4, 4-Chloro-3-nitrophenylboronic acid 151169-69-6 151169-71-0 159896-15-8 178320-31-5 RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (chemiluminescent reactions enhancement with organoboron and phenolic or amine compds.) 1679-18-1 HCAPLUS RNBoronic acid, (4-chlorophenyl) - (9CI) (CA INDEX NAME) CN

RN 5122-94-1 HCAPLUS CN Boronic acid, [1,1'-biphenyl]-4-yl- (9CI) (CA INDEX NAME)

RN 5122-99-6 HCAPLUS CN Boronic acid, (4-iodophenyl) - (9CI) (CA INDEX NAME)

RN 5467-74-3 HCAPLUS CN Boronic acid, (4-bromophenyl)- (9CI) (CA INDEX NAME)

RN 17865-11-1 HCAPLUS
CN Boronic acid, [4-(trimethylsilyl)phenyl]- (9CI) (CA INDEX NAME)

RN 96983-22-1 HCAPLUS CN Boronic acid, (4-bromophenyl)-, dibutyl ester (9CI) (CA INDEX NAME)

RN 151169-66-3 HCAPLUS CN Boronic acid, [2-hydroxy-5-[[3-(trifluoromethyl)phenyl]azo]phenyl]- (9CI) (CA INDEX NAME)

RN 151169-67-4 HCAPLUS CN Boronic acid, (4-chloro-3-nitrophenyl)- (9CI) (CA INDEX NAME)

RN 151169-69-6 HCAPLUS

CN Boronic acid, (4'-bromo[1,1'-biphenyl]-4-yl)-, dibutyl ester (9CI) (CA INDEX NAME)

RN 151169-71-0 HCAPLUS

CN Boronic acid, [4-[(4-chlorophenyl)amino]phenyl]- (9CI) (CA INDEX NAME)

RN 159896-15-8 HCAPLUS

CN 2-Propenoic acid, 3-(4-boronophenyl)-, (2E)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

RN 178320-31-5 HCAPLUS

CN Benzoic acid, 4-[(3-borono-4-hydroxyphenyl)azo]- (9CI) (CA INDEX NAME)

L28 ANSWER 30 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: 1996:148680 HCAPLUS

DOCUMENT NUMBER: 124:254001

TITLE: Comparison, of 5-hydroxy-2,3-dihydrophthalazine-1,4-

dione and luminol as co-substrates for detection of

horseradish peroxidase in enhanced

chemiluminescent reactions

AUTHOR(S): Kricka, Larry J.; Ji, Xiaoying; Thorpe, Gary H. G.;

Edwards, Brooks; Voyta, John; Bronstein, Irena

CORPORATE SOURCE: Dep. Pathology Laboratory Medicine, Univ.

Pennsylvania, Philadelphia, PA, 19104, USA Journal of Immunoassay (1996), 17(1), 67-83

CODEN: JOUIDK; ISSN: 0197-1522

PUBLISHER: Dekker
DOCUMENT TYPE: Journal
LANGUAGE: English

The utility of 5-hydroxy-2,3-dihydrophthalazine-1,4-dione (HDP) as a co-substrate for the chemiluminescent detection of horseradish peroxidase was assessed. Several substituted aryl boronic acid derivs. (4-Ph, 4-iodo) acted as potent enhancers of the peroxidase catalyzed reaction. Addition of chelating agents (EDTA) and surfactants (Tween-20 and [poly(vinylbenzyl)tributylphosphonium chloride-poly (vinylbenzyl)trioctylphosphonium chloride copolymer]) modulated background light emission and the intensity and duration of the signal from both HDP and luminol. However, HDP was found to be inferior to luminol in the peroxidase assay. Comparative studies revealed that at 500 amol of peroxidase the S/B was ten-fold higher using a com. luminol-based signal reagent as compared with an HDP-EDTA-Tween-20 reagent (S/B t = 0 min 21.8 vs. 1.7, S/B t = 10 min 17.8 vs. 2.0).

CC 7-1 (Enzymes)

SOURCE:

ST peroxidase chemiluminescent detn hydroxydihydrophthalazinedione luminol

IT 9003-99-0, Peroxidase

RL: ANT (Analyte); ANST (Analytical study)
(comparison, of 5-hydroxy-2,3-dihydrophthalazine-1,4-dione and luminol as co-substrates for detection of horseradish peroxidase in enhanced chemiluminescent reactions)

TT 521-31-3, Luminol 7600-08-0, 5-Hydroxy-2,3-dihydrophthalazine-1,4-dione RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (comparison, of 5-hydroxy-2,3-dihydrophthalazine-1,4-dione and luminol as co-substrates for detection of horseradish peroxidase in enhanced chemiluminescent reactions)

IT 5122-94-1 159896-15-8

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (enhancer; comparison, of 5-hydroxy-2,3-dihydrophthalazine-1,4-dione and luminol as co-substrates for detection of horseradish peroxidase in enhanced chemiluminescent reactions)

IT 60-00-4, EDTA, uses 9005-64-5, Tween-20

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (stabilizer; comparison, of 5-hydroxy-2,3-dihydrophthalazine-1,4-dione and luminol as co-substrates for detection of horseradish peroxidase in enhanced chemiluminescent reactions)

IT 5122-94-1 159896-15-8

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (enhancer; comparison, of 5-hydroxy-2,3-dihydrophthalazine-1,4-dione and luminol as co-substrates for detection of horseradish peroxidase in enhanced chemiluminescent reactions)

RN 5122-94-1 HCAPLUS

CN Boronic acid, [1,1'-biphenyl]-4-yl- (9CI) (CA INDEX NAME)

RN 159896-15-8 HCAPLUS

CN 2-Propenoic acid, 3-(4-boronophenyl)-, (2E)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

L28 ANSWER 31 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1995:702883 HCAPLUS

DOCUMENT NUMBER:

123:137837

TITLE:

New enhancers for the chemiluminescent peroxidase catalyzed chemiluminescent

oxidation of pyrogallol and purpurogallin

AUTHOR(S): CORPORATE SOURCE: Nozaki, Osamu; Ji, Xiaoying; Kricka, Larry J. Department of Pathology and Laboratory Medicine,

University of Pennsylvania, Philadelphia, PA,

19104-4283, USA

SOURCE:

Journal of Bioluminescence and Chemiluminescence

(1995), 10(3), 151-6

CODEN: JBCHE7; ISSN: 0884-3996

DOCUMENT TYPE:

Journal

LANGUAGE:

English

AB The effects of various boronate compds., 4-biphenylboronic acid, 4-bromobenzene-boronic acid, trans-4-(3-propionic acid) phenylboronic acid and 4-iodophenylboronic acid, on the horseradish peroxidase (HRP) catalyzed chemiluminescent oxidation of pyrogallol and purpurogallin by peroxide were investigated. Trans-4-(3-Propionic

acid) phenylboronic acid produced a 13.7-fold enhancement in the peak light emission from the **chemiluminescent** HRP catalyzed pyrogallol reaction (detection limit for HRP < 1.25 fmol). At low enhancer concentration

single peak of light emission was observed and as the enhancer concentration increased the time to peak light emission became progressively longer. The **chemiluminescence** showed two peaks at higher concns. (>54.3 $\mu\text{mol/L})$ and the individual peak times depended upon the concentration of the enhancer. All of the boronates enhanced peak light emission in the **chemiluminescent** HRP catalyzed purpurogallin reaction. 4-Biphenylboronic acid was the most effective and it enhanced peak light emission 314-fold. The practical detection limit for HRP (Type VIA) using this enhancer was 4.18 pmol (peak emission at 20 min). This compound also enhanced peak light emission 232-fold from a **chemiluminescent** HRP-purpurogallin reaction in which mol. oxygen replaced peroxide as the oxidant.

CC 9-5 (Biochemical Methods)

ST enhancer **chemiluminescent** peroxidase oxidn pyrogallol purpurogallin

IT Oxidation

(chemiluminescent; new enhancers for chemiluminescent
peroxidase catalyzed chemiluminescent oxidation of pyrogallol
and purpurogallin)

IT 9003-99-0, Peroxidase

RL: CAT (Catalyst use); USES (Uses)

(new enhancers for **chemiluminescent** peroxidase catalyzed **chemiluminescent** oxidation of pyrogallol and purpurogallin)

IT 5122-94-1, 4-Biphenylboronic acid 5122-99-6

5467-74-3, 4-Bromobenzene-boronic acid 166316-48-9

RL: NUU (Other use, unclassified); USES (Uses)

(new enhancers for **chemiluminescent** peroxidase catalyzed **chemiluminescent** oxidation of pyrogallol and purpurogallin)

IT 87-66-1, Pyrogallol 569-77-7, Purpurogallin

RL: RCT (Reactant); RACT (Reactant or reagent)

(new enhancers for **chemiluminescent** peroxidase catalyzed **chemiluminescent** oxidation of pyrogallol and purpurogallin)

IT 5122-94-1, 4-Biphenylboronic acid 5122-99-6

5467-74-3, 4-Bromobenzene-boronic acid 166316-48-9

RL: NUU (Other use, unclassified); USES (Uses)

(new enhancers for **chemiluminescent** peroxidase catalyzed **chemiluminescent** oxidation of pyrogallol and purpurogallin)

RN 5122-94-1 HCAPLUS

CN Boronic acid, [1,1'-biphenyl]-4-yl- (9CI) (CA INDEX NAME)

RN 5122-99-6 HCAPLUS

CN Boronic acid, (4-iodophenyl) - (9CI) (CA INDEX NAME)

RN5467-74-3 HCAPLUS

Boronic acid, (4-bromophenyl) - (9CI) (CA INDEX NAME) CN

RΝ 166316-48-9 HCAPLUS

CN Benzenepropanoic acid, 4-borono- (9CI) (CA INDEX NAME)

L28 ANSWER 32 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1995:485390 HCAPLUS

DOCUMENT NUMBER:

123:50528

TITLE:

4-Phenylylboronic acid: a new type of enhancer for the

horseradish peroxidase catalyzed

chemiluminescent oxidation of luminol

AUTHOR (S):

Kricka, Larry J.; Ji, Xiaoying

CORPORATE SOURCE:

Department of Pathology and Laboratory Medicine, University of Pennsylvania, Philadelphia, PA, 19104,

SOURCE:

Journal of Bioluminescence and Chemiluminescence

(1995), 10(1), 49-54

CODEN: JBCHE7; ISSN: 0884-3996

DOCUMENT TYPE:

Journal

LANGUAGE:

English

4-Phenylylboronic acid enhances the light emission from the horseradish peroxidase-catalyzed oxidation of luminol by hydrogen peroxide. Optimization studies showed that the greatest enhancement was obtained using micromolar concns. of the new enhancer. The largest degree of enhancement was found with the basic isoenzyme of horseradish peroxidase (Type VIA), and lesser degrees of enhancement were obtained with Type VII and Type IX horseradish

peroxidase. The enhancer was also effective in the peroxidase-catalyzed oxidation of isoluminol by peroxide.

CC 7-1 (Enzymes)

ST phenylylboronate enhancer peroxidase chemiluminescence luminol

IT 9003-99-0, Peroxidase

RL: ANT (Analyte); ANST (Analytical study)

(4-phenylylboronic acid as new type of enhancer for horseradish peroxidase-catalyzed **chemiluminescent** oxidation of luminol)

IT 5122-94-1

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (4-phenylylboronic acid as new type of enhancer for horseradish peroxidase-catalyzed chemiluminescent oxidation of luminol)

IT 521-31-3, Luminol 3682-14-2, Isoluminol

RL: RCT (Reactant); RACT (Reactant or reagent)

(4-phenylylboronic acid as new type of enhancer for horseradish peroxidase-catalyzed **chemiluminescent** oxidation of luminol)

IT 5122-94-1

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (4-phenylylboronic acid as new type of enhancer for horseradish peroxidase-catalyzed chemiluminescent oxidation of luminol)

RN 5122-94-1 HCAPLUS

CN Boronic acid, [1,1'-biphenyl]-4-yl- (9CI) (CA INDEX NAME)

L28 ANSWER 33 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1995:264714 HCAPLUS

DOCUMENT NUMBER:

122:27259

TITLE:

Enhancers for chemiluminescent reactions

INVENTOR(S):

Kricka, Larry Jan

PATENT ASSIGNEE(S):

British Technology Group Ltd., UK

SOURCE:

Brit. UK Pat. Appl., 47 pp.

CODEN: BAXXDU

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

DAMINIM NO	KIND	DATE	APPLICATION NO.	DATE
PATENT NO.	KTND	DAIL	APPLICATION NO.	DAIL
GB 2276721	A1	19941005	GB 1994-6587	19940331
GB 2276721	B2	19960821	•	
WO 9423060	A2	19941013	WO 1994-GB700	19940331
WO 9423060	A 3	19941124		
W: AU, CA, FI,	JP			
RW: AT, BE, CH,	DE, DK	, ES, FR, G	B, GR, IE, IT, LU, MC,	NL, PT, SE
CA 2157592	AA	19941013	CA 1994-2157592	19940331
AU 9463827	A1	19941024	AU 1994-63827	19940331
EP 692033	A1	19960117	EP 1994-911267	19940331

```
EP 692033
                          B1
                                19980311
         R: BE, CH, DE, ES, FR, IE, IT, LI, NL, SE
     JP 08508342
                         T2
                                19960903
                                            JP 1994-521849
                                                                   19940331
                                            US 1995-478336
     US 5629168
                                19970513
                         Α
                                                                   19950607
     FI 9504653
                          Α
                                19950929
                                            FI 1995-4653
                                                                   19950929
PRIORITY APPLN. INFO.:
                                            GB 1993-6888
                                                                   19930401
                                            US 1992-833153
                                                                   19920210
                                            US 1992-925689
                                                                   19920807
                                            US 1993-157504
                                                                   19931126
                                            WO 1994-GB700
                                                                   19940331
     In enhanced chemiluminescent (ECL) reactions of a fused aromatic
AB
     diacyl cyclic hydrazide such as luminol, a peroxidase enzyme catalyst, an
     oxidant such as hydrogen peroxide and an enhancer, it has been found
     advantageous to use a combination of an organoboron enhancer such as
     4-biphenylboronic acid with a non boron-containing enhancer, especially a
phenolic
     or aromatic amine enhancer, particularly 4-iodophenol ECL reactions are
     useful in diagnostic assay.
IC
     ICM G01N021-76
     ICS C12Q001-28; G01N033-58
CC
     9-5 (Biochemical Methods)
     chemiluminescent enhancer arom diacyl cyclic hydrazide;
ST
     peroxidase enzyme oxidant chemiluminescent reaction enhancer
     Blood analysis
IT
       Luminescence, chemi-
        (enhancers for chemiluminescent reactions)
IT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (cyclic, fused aromatic diacyl; enhancers for chemiluminescent
        reactions)
IT
     9002-71-5, Thyrotropin
     RL: ANT (Analyte); ANST (Analytical study)
        (enhancers for chemiluminescent reactions)
     9003-99-0, Peroxidase
TT
     RL: CAT (Catalyst use); USES (Uses)
        (enhancers for chemiluminescent reactions)
\mathbf{IT}
     92-04-6, 2-Chloro-4-Phenylphenol 92-69-3, 4-Phenylphenol
                                                                  95-77-2,
     3,4-Dichlorophenol 98-54-4 101-53-1, 4-Benzylphenol 103-90-2,
     4-Acetamidophenol
                        104-94-9
                                  106-41-2, 4-Bromophenol
                                                             106-44-5, uses
     106-48-9, 4-Chlorophenol 120-83-2, 2,4-Dichlorophenol
                                                               135-19-3,
     2-Naphthalenol, uses
                          540-38-5, 4-Iodophenol
                                                     573-97-7
                                                                637-89-8
     831-82-3, 4-Phenoxyphenol 1634-82-8 1679-18-1,
     4-Chlorophenylboronic acid
                                1689-82-3, 4-(Phenylazo) phenol
                                                                   1965-09-9
                            3558-83-6, 4-(4'-Hydroxyphenyl)benzophenone
     2848-16-0 2975-55-5
     3964-56-5, 4-Bromo-2-chlorophenol 5122-94-1, 4-Biphenylboronic
     acid 5122-99-6 5467-74-3 7400-08-0,
     4-Hydroxycinnamic acid 10041-02-8
                                         15015-57-3, 4-Hydroxyphenyl
     disulfide
               15231-91-1
                             16239-18-2 17865-11-1,
     4-(Trimethylsilyl)benzeneboronic acid 23567-67-1
                                                          23795-02-0, Ethyl
     3-(4-hydroxyphenyl)propionate 34314-06-2, Tetramethylbenzidine
     83474-09-3
                 92681-33-9 96983-22-1 144774-20-9 144774-21-0
     150296-58-5 151169-66-3 151169-67-4,
     4-Chloro-3-nitrophenylboronic acid 151169-69-6
                                                     151169-70-9
     151169-71-0 159896-15-8
    RL: MOA (Modifier or additive use); USES (Uses)
        (enhancers for chemiluminescent reactions)
IT
    521-31-3, Luminol
                       1445-69-8
                                   7722-84-1, Hydrogen peroxide, reactions
    RL: RCT (Reactant); RACT (Reactant or reagent)
        (enhancers for chemiluminescent reactions)
```

IT 1679-18-1, 4-Chlorophenylboronic acid 5122-94-1,
4-Biphenylboronic acid 5122-99-6 5467-74-3
17865-11-1, 4-(Trimethylsilyl)benzeneboronic acid
96983-22-1 151169-66-3 151169-67-4,
4-Chloro-3-nitrophenylboronic acid 151169-69-6
151169-71-0 159896-15-8
RL: MOA (Modifier or additive use); USES (Uses)
(enhancers for chemiluminescent reactions)
RN 1679-18-1 HCAPLUS
CN Boronic acid, (4-chlorophenyl) - (9CI) (CA INDEX NAME)

RN 5122-94-1 HCAPLUS CN Boronic acid, [1,1'-biphenyl]-4-yl- (9CI) (CA INDEX NAME)

RN 5122-99-6 HCAPLUS CN Boronic acid, (4-iodophenyl)- (9CI) (CA INDEX NAME)

RN 5467-74-3 HCAPLUS CN Boronic acid, (4-bromophenyl)- (9CI) (CA INDEX NAME)

RN 17865-11-1 HCAPLUS

CN Boronic acid, [4-(trimethylsilyl)phenyl] - (9CI) (CA INDEX NAME)

RN 96983-22-1 HCAPLUS

CN Boronic acid, (4-bromophenyl)-, dibutyl ester (9CI) (CA INDEX NAME)

RN 151169-66-3 HCAPLUS

CN Boronic acid, [2-hydroxy-5-[[3-(trifluoromethyl)phenyl]azo]phenyl]- (9CI) (CA INDEX NAME)

RN 151169-67-4 HCAPLUS

CN Boronic acid, (4-chloro-3-nitrophenyl) - (9CI) (CA INDEX NAME)

RN 151169-69-6 HCAPLUS

CN Boronic acid, (4'-bromo[1,1'-biphenyl]-4-yl)-, dibutyl ester (9CI) (CA INDEX NAME)

RN 151169-71-0 HCAPLUS

CN Boronic acid, [4-[(4-chlorophenyl)amino]phenyl]- (9CI) (CA INDEX NAME)

RN 159896-15-8 HCAPLUS

CN 2-Propenoic acid, 3-(4-boronophenyl)-, (2E)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

L28 ANSWER 34 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1994:675102 HCAPLUS

DOCUMENT NUMBER:

121:275102

TITLE:

Rate constants for reactions of horseradish peroxidase compounds I and II with 4-substituted arylboronic

acids

AUTHOR (S):

SOURCE:

Sun, Weimei; Ji, Xiaoying; Kricka, Larry J.; Dunford,

H. Brian

CORPORATE SOURCE:

Dep. Chem., Univ. Alberta, Edmonton, AB, T6G 2G2, Can. Canadian Journal of Chemistry (1994), 72(10), 2159-62

CODEN: CJCHAG; ISSN: 0008-4042

National Research Council of Canada

PUBLISHER:

Journal

DOCUMENT TYPE: LANGUAGE:

English

The rate consts. for the reactions of horseradish peroxidase compound I (k1) and compound II (k2) with three 4-substituted arylboronic acids, which enhance chemiluminescence in the horseradish peroxidase catalyzed oxidation of luminol by hydrogen peroxide, were determined at pH 8.6, total ionic strength 0.11 M, using stopped-flow kinetic measurements. For comparison, the rate consts. of the reactions of 4-iodophenol with compds. I and II were also determined under the same exptl. conditions. The three arylboronic acid derivs. and their rate consts. are: 4-biphenylboronic acid, $k1 = (1.21 \pm 0.08) + 106 M-1 s-1, k2 = (4.6 \pm 0.2)$ + 105 M-1 s-1; 4-bromophenylboronic acid, $k1 = (5.5 \pm 0.2)$ + 104 M-1 s-1, $k2 = (3.6 \pm 0.2) + 104 \text{ M}-1 \text{ s}-1$; and 4-iodophenylboronic acid, $k1 = (1.1 \pm 0.2) + 105 \text{ M-1 s-1}, k2 =$ $(1.3 \pm 0.1) + 104 \text{ M-1 s-1}$. 4-Biphenylboronic acid, which shows comparable luminescent enhancement to 4-iodophenol, has the highest reactivity in the reduction of both compds. I and II among the three arylboronic acid derivs. tested.

7-3 (Enzymes) CC

IT 5122-94-1, 4-Biphenylboronic acid 5122-99-6, Boronic acid, (4-iodophenyl) - 5467-74-3, 4-Bromophenylboronic acid 9003-99-0, Peroxidase

RL: RCT (Reactant); RACT (Reactant or reagent)

(rate consts. for reactions of horseradish peroxidase compds. I and II with 4-substituted arylboronic acids)

IT5122-94-1, 4-Biphenylboronic acid 5122-99-6, Boronic

acid, (4-iodophenyl) - 5467-74-3, 4-Bromophenylboronic acid

RL: RCT (Reactant); RACT (Reactant or reagent)

(rate consts. for reactions of horseradish peroxidase compds. I and II with 4-substituted arylboronic acids)

5122-94-1 HCAPLUS RN

CN Boronic acid, [1,1'-biphenyl]-4-yl- (9CI) (CA INDEX NAME)

RN5122-99-6 HCAPLUS

Boronic acid, (4-iodophenyl) - (9CI) (CA INDEX NAME) CN

RN 5467-74-3 HCAPLUS

CN Boronic acid, (4-bromophenyl) - (9CI) (CA INDEX NAME)

L28 ANSWER 35 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER:

1993:644979 HCAPLUS

DOCUMENT NUMBER:

119:244979

TITLE:

Organoboron compounds as chemiluminescence

enhancers

INVENTOR(S):

Kricka, Larry Jan

PATENT ASSIGNEE(S):

British Technology Group Ltd., UK

SOURCE:

PCT Int. Appl., 36 pp. CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

English

1

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

	PATENT NO.							DATE											
	WO 9316195						WO 1993-GB271												
	W: AU, CA, FI,																		
		RW:	AT,	BE,	CH,	DE,	DK,	, ES,	FR,	GB,	GF	R, IE	Ξ, Ξ	IT,	LU,	MC,	NL,	PT,	se
	AU	9334	598			A1		1993	0903		AU	1993	3 - 3 4	4598	В		1	9930	210
AU 665234					B2		1995	1221											
	GB	2265	459			A1		1993	0929		GB	1993	3-2	573			1	9930	210
	GB	2265	459			B2		1995	1206										
	ZA	9300	922			Α		1994	0810		ZA	1993	3 - 92	22			1	9930	210
	ΕP	6260	16			A1		1994	1130		EΡ	1993	3-91	032	59		1	9930	210
	ΕP	6260	16			В1		1998	0603										
		R:	BE,	CH,	DE,	ES,	FR	, GB,	ΙE,	IT,	L	I, NI	٠, ١	SE					
	JP	0750						1995	0427		JP	1993	3-5	1389	98		1	9930	210
		2121																9930	210
		9403																9940	810
PRIO		APP										1992						9920	210
											GB	1992	2-1	6784	4		1	9920	807
											WO	1993	3 – GI	B27	1		1	9930	210
OTHER SOURCE(S):						MAR	TAG	119:	2449	79									
3.7	~	1.				т Г1		II D	1 4		CCI	14 5		E 1 /	7770	cu2	or	(OD)	2 -

AB Organoboron compds. I [R = H, Bu, 4'-ClC6H4, 3',5'-Cl2C6H3, or (OR)2 =

```
O,O-propylene; W = H, OH, Me, MeO, Cl, etc.; X = H, Me, Cl, NH2, NO2,
    etc.; Y = H, Me, CO2H, Cl, etc.; Z = H, 5- or 6-Cl, 5-Br, etc.] increase
    the light output and/or signal from a chemiluminescence reaction
     involving a dihydrophthalazinedione, a peroxidase, and an oxidant.
    Compds. increasing the signal:background ratio by >25% were selected as
    chemiluminescence enhancers for use in diagnostic immunoassays.
    ICM C12Q001-28
IC
CC
    9-10 (Biochemical Methods)
    Section cross-reference(s): 79, 80
    organoboron compd chemiluminescence enhancer
ST
    Immunoassay
IT
       Spectrochemical analysis
        (chemiluminescence, signal enhancers for, organoboron compds.
        as)
    98-80-6, Phenylboronic acid 1679-18-1
                                             2848-16-0
TT
    4151-80-8 4406-77-3 5122-94-1
    5122-99-6, p-Iodophenylboronic acid 5467-74-3,
    p-Bromophenylboronic acid 5720-05-8 13331-27-6,
     3-Nitrophenylboronic acid 13922-41-3
                 16986-25-7 17865-11-1, 4-
    16419-60-6
                                           24493-75-2 30418-59-8
     (Trimethylsilyl) benzeneboronic acid
    51067-38-0 63503-60-6 80500-28-3
                             149196-90-7 151169-66-3
    89694-45-1 96983-22-1
    151169-67-4, 4-Chloro-3-nitrophenylboronic acid
    151169-68-5 151169-69-6
                               151169-70-9
    151169-71-0 151169-72-1
                               151169-73-2
                               151169-76-5
    151169-74-3 151169-75-4
    151196-37-1
                  151233-30-6
    RL: ANST (Analytical study)
        (as chemiluminescence enhancer)
    9003-99-0, Peroxidase
IT
    RL: ANST (Analytical study)
        (chemiluminescent reaction involving dihydrophthalazinedione
        and oxidant and, organoboron compds. as chemiluminescence
        enhancers for)
TT
    7722-84-1, Hydrogen peroxide, reactions
    RL: RCT (Reactant); RACT (Reactant or reagent)
        (chemiluminescent reaction involving dihydrophthalazinedione
        and peroxidase and, organoboron compds. as chemiluminescence
        enhancers for)
    521-31-3, Luminol
IT
    RL: ANST (Analytical study)
        (chemiluminescent reaction involving oxidant and peroxidase
        and, organoboron compds. as chemiluminescence enhancers for)
IT
    1445-69-8
    RL: ANST (Analytical study)
        (chemiluminescent reaction involving peroxidase and oxidant
        and, organoboron compds. as chemiluminescence enhancers for)
    98-80-6, Phenylboronic acid 1679-18-1 4151-80-8
IT
    4406-77-3 5122-94-1 5122-99-6,
    p-Iodophenylboronic acid 5467-74-3, p-Bromophenylboronic acid
    5720-05-8 13331-27-6, 3-Nitrophenylboronic acid
    13922-41-3 16419-60-6 17865-11-1,
    4-(Trimethylsilyl)benzeneboronic acid 30418-59-8
    51067-38-0 63503-60-6 80500-28-3
    89694-45-1 96983-22-1 151169-66-3
    151169-67-4, 4-Chloro-3-nitrophenylboronic acid
    151169-68-5 151169-69-6 151169-71-0
    151169-72-1 151169-74-3 151169-75-4
```

151196-37-1

RL: ANST (Analytical study)

(as chemiluminescence enhancer)

RN 98-80-6 HCAPLUS

CN Boronic acid, phenyl- (9CI) (CA INDEX NAME)

RN 1679-18-1 HCAPLUS

CN Boronic acid, (4-chlorophenyl) - (9CI) (CA INDEX NAME)

RN 4151-80-8 HCAPLUS

CN Boronic acid, [1,1'-biphenyl]-4,4'-diylbis- (9CI) (CA INDEX NAME)

RN 4406-77-3 HCAPLUS

CN 1,3,2-Dioxaborinane, 2-phenyl- (9CI) (CA INDEX NAME)

RN 5122-94-1 HCAPLUS

CN Boronic acid, [1,1'-biphenyl]-4-yl- (9CI) (CA INDEX NAME)

RN 5122-99-6 HCAPLUS CN Boronic acid, (4-iodophenyl)- (9CI) (CA INDEX NAME)

RN 5467-74-3 HCAPLUS CN Boronic acid, (4-bromophenyl)- (9CI) (CA INDEX NAME)

RN 5720-05-8 HCAPLUS CN Boronic acid, (4-methylphenyl)- (9CI) (CA INDEX NAME)

RN 13331-27-6 HCAPLUS CN Boronic acid, (3-nitrophenyl)- (9CI) (CA INDEX NAME)

RN 13922-41-3 HCAPLUS

CN Boronic acid, 1-naphthalenyl- (9CI) (CA INDEX NAME)

RN 16419-60-6 HCAPLUS

CN Boronic acid, (2-methylphenyl) - (9CI) (CA INDEX NAME)

RN 17865-11-1 HCAPLUS

CN Boronic acid, [4-(trimethylsilyl)phenyl]- (9CI) (CA INDEX NAME)

RN 30418-59-8 HCAPLUS

CN Boronic acid, (3-aminophenyl) - (9CI) (CA INDEX NAME)

RN 51067-38-0 HCAPLUS

CN Boronic acid, (4-phenoxyphenyl) - (9CI) (CA INDEX NAME)

RN 63503-60-6 HCAPLUS

CN Boronic acid, (3-chlorophenyl) - (9CI) (CA INDEX NAME)

RN 80500-28-3 HCAPLUS

CN Benzoic acid, 4-borono-2-nitro- (6CI, 9CI) (CA INDEX NAME)

RN 89694-45-1 HCAPLUS

CN Boronic acid, (5-bromo-2-methoxyphenyl) - (9CI) (CA INDEX NAME)

RN 96983-22-1 HCAPLUS CN Boronic acid, (4-bromophenyl)-, dibutyl ester (9CI) (CA INDEX NAME)

RN 151169-66-3 HCAPLUS
CN Boronic acid, [2-hydroxy-5-[[3-(trifluoromethyl)phenyl]azo]phenyl]- (9CI)
(CA INDEX NAME)

RN 151169-67-4 HCAPLUS CN Boronic acid, (4-chloro-3-nitrophenyl)- (9CI) (CA INDEX NAME)

RN 151169-68-5 HCAPLUS CN 2-Propenoic acid, 3-(4-boronophenyl)- (9CI) (CA INDEX NAME)

$$_{
m HO-B}^{
m OH}$$
 $_{
m CH-CO_2H}^{
m CH-CO_2H}$

RN 151169-69-6 HCAPLUS

CN Boronic acid, (4'-bromo[1,1'-biphenyl]-4-yl)-, dibutyl ester (9CI) (CA INDEX NAME)

RN 151169-71-0 HCAPLUS

CN Boronic acid, [4-[(4-chlorophenyl)amino]phenyl]- (9CI) (CA INDEX NAME)

RN 151169-72-1 HCAPLUS

CN Boronic acid, (7-hydroxy-2-naphthalenyl) - (9CI) (CA INDEX NAME)

RN 151169-74-3 HCAPLUS

CN Boronic acid, (2,3-dichlorophenyl) - (9CI) (CA INDEX NAME)

RN 151169-75-4 HCAPLUS

CNBoronic acid, (3,4-dichlorophenyl) - (9CI) (CA INDEX NAME)

RN151196-37-1 HCAPLUS

Boronic acid, (3-amino-2,4,6-trichlorophenyl) - (9CI) (CA INDEX NAME) CN

$$C1$$
 $C1$ $B-OH$ $C1$ OH

L28 ANSWER 36 OF 36 HCAPLUS COPYRIGHT 2004 ACS on STN 1990:213569 HCAPLUS

ACCESSION NUMBER:

112:213569 DOCUMENT NUMBER:

Tridentate conjugates for competitive immunoassays TITLE:

INVENTOR(S): Oh, Chan S.; Sternberg, James C. Beckman Instruments, Inc., USA PATENT ASSIGNEE(S):

Eur. Pat. Appl., 40 pp. SOURCE:

CODEN: EPXXDW

DOCUMENT TYPE: Patent English LANGUAGE:

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.					KIN	D C	DATE		APPLICATION NO.	DATE	
							_				
	EP	3103	61			A2		1989	0405	EP 1988-309002	19880929
	EP	3103	61			A3		1989	0524		
		R:	AT,	BE,	CH,	DE,	ES,	FR,	GB,	GR, IT, LI, LU, NL, SE	
	WO	8903	041			A2		1989	0406	WO 1988-US3368	19880930
	WO	8903	041			А3		1989	0420	,	

```
W: AU, JP
        RW: AT, BE, CH, DE, FR, GB, IT, LU, NL, SE
                             19890418
                                        AU 1988-26056
                                                              19880930
    AU 8826056
                       A1
    AU 623352
                       B2
                             19920514
    US 5168057
                       Α
                             19921201 US 1991-768118
                                                              19910930
    JP 06222058
                       A2 19940812
                                        JP 1992-225325
                                                              19920731
    JP 2627124
                       B2 19970702
    US 5661019
                       A 19970826
                                        US 1995-410014
                                                              19950322
    US 5851778
                       Α
                             19981222
                                        US 1997-832143
                                                              19970402
PRIORITY APPLN. INFO.:
                                         US 1987-103093
                                                              19870930
                                         WO 1988-US3368
                                                              19880930
                                         US 1991-768118
                                                              19910930
                                         US 1992-911827
                                                              19920710
                                         US 1995-410014
                                                              19950322
```

AB A tridentate conjugate for competitive immunoassays has 3 chemical moieties, or tridentate members, attached through an appropriate spacer moiety. At least 2 of the tridentate members are relatively small mols. (e.g. ligands, haptens), usually .ltorsim.7000 daltons. The particular appropriate spacer moiety selected for a tridentate imparts certain steric properties to the tridentate conjugate. In 1 embodiment, the binding of a macromol. specific binding partner to one of the tridentate members sterically inhibits the binding of a different macromol. to another tridentate member. In another embodiment, the binding of a 1st tridentate member to a macromol. restricts the subsequent binding of a 2nd tridentate member to a proximate location on the same macromol. Thus, a biotin-theophylline-lysine conjugate (preparation described) was reacted with DNP-bis(aminocaproic acid) N-hydroxysuccinimide ester (preparation described) to form a biotin-theophylline-DNP conjugate. Theophylline amine (I) was determined in a nephelometric inhibition immunoassay by mixing the conjugate with anti-theophylline monoclonal antibody, anti-DNP antibody, avidin, and samples containing the analyte. Free I competed with theophylline in the conjugate for the anti-theophylline monoclonal antibody. Increasing concns. of I resulted in an increased nephelometric signal.

IC ICM G01N033-531

IT

ICS G01N033-94; G01N033-532

CC 9-10 (Biochemical Methods)

IT Luminescent substances

(chemi-, conjugates with haptens and macromols., tridentate, for competitive immunoassays)

IT Immunochemical analysis

(chemiluminescence energy-transfer immunoassay, tridentate conjugates for, preparation of)

81-88-9D, tridentate conjugates with haptens and macromols. 98-80-6D, Phenyl boronic acid, tridentate conjugates with haptens 1445-69-8D, tridentate conjugates with haptens and and macromols. 2321-07-5D, Fluorescein, tridentate conjugates with haptens macromols. 7440-31-5D, Tin, protoporphyrin complexes, tridentate and macromols. conjugates with haptens and macromols. 7440-66-6D, Zinc, protoporphyrin complexes, tridentate conjugates with haptens and macromols. 9003-99-0D, Peroxidase, tridentate conjugates with haptens and macromols. 109392-90-7D, tridentate conjugates with haptens and macromols. 9001-37-0D, Glucose oxidase, tridentate conjugates with haptens and 9001-40-5D, Glucose-6-phosphate dehydrogenase, tridentate conjugates with haptens and macromols. 9001-51-8D, Hexokinase, tridentate conjugates with haptens and macromols. RL: ANST (Analytical study)

(for competitive immunoassays)

IT 98-80-6D, Phenyl boronic acid, tridentate conjugates with haptens and macromols. RL: ANST (Analytical study)
(for competitive immunoassays)
RN 98-80-6 HCAPLUS
CN Boronic acid, phenyl- (9CI) (CA INDEX NAME)

Рh | HO— В— ОН